β_3 -ADRENORECEPTOR AGONISTS, AGONIST COMPOSITIONS AND METHODS OF USING

CROSS-REFERENCE TO RELATED APPLICATION

The present application is a continuation-in-part application of U.S. Provisional Patent Application Serial No. 60/061,152 filed by the same inventors on September 30, 1997, the entire disclosure of which is incorporated herein by reference.

BACKGROUND

TECHNICAL FIELD

The present invention relates to the field of β_3 -Adrenoreceptor agonists and to methods of their preparation, formulation and use to stimulate, regulate and modulate metabolism of fats in adipose tissues in animals, particularly humans and other mammals. More particularly, the present invention relates to the field of treating obesity and overweight conditions in animals, particularly humans and other mammals and associated effects of conditions associated with obesity and overweight, including Type II diabetes mellitus (non-insulin dependent diabetes), insulin resistance, glucose intolerance, hypothyroidism, morbid obesity, and the like.

PRIOR ART

It was long thought that obesity was a consequence of self-indulgence and undisciplined behavior. Obesity was seen as evidence of gluttony, through a lack of will or capacity for self-discipline. The overweight have been disparaged, and thinness has been celebrated. Indeed, the perception of thinness as a major aspect of human beauty and attractiveness has become endemic in modern culture, and overweight conditions and obesity has increasingly grown to be an unacceptable condition for social reasons.

Masked by these cultural icons are the hard medical facts: for many individuals, a tendency to overweight and even obesity are often symptoms of organic disease or disorders of the metabolism, associated with serious and even

life-threatening conditions. In medical economic terms alone, the costs attributable to overweight and obesity are staggeringly high.

A wide variety of approaches to the alleviation of obesity have ebbed and flowed though modern culture, ranging from a diverse collection of dietary strategies, to drugs, to surgical interventions, to hypnosis. All have met with indifferent success at best. Some have proved to be outright quackery. Others have proved to be effective only for the short-term, with loss of effectiveness over time. Still others have proved to be generally or at least partially successful so long as the regimen is sustained, but long term compliance is difficult to attain and in some cases has proved hazardous to other aspects of health and well-being. Some surgical procedures have had some successes, but as with any invasive procedures, there are risks. Some approaches to weight loss and control, in the extreme, lead to conditions which are themselves pathological, such as bulimia and anorexia nervosa. Other effects are less extreme, but still highly undesirable, such as amennorhea, vitamin and essential nutrient deficiencies, and the like.

A great deal of the difficulty in the art and practice of obesity and overweight management has been a consequence of attention focused on the control of appetite, and reducing the amount of food intake. It has long been the belief of many that only by the control of caloric intake is it possible to regulate body weight and fat deposition and utilization. Since appetite is controlled and regulated in the brain, brain pharmacology and the alteration of brain chemistry has been a primary focus of weight regulation and control efforts. Such approaches have led to addictions to appetite suppressants, to primary pulmonary myopathy, cardiac valve damage, and to reports of serotonin disruptions and disorders and psychotic episodes among users. Morbities and mortalities have been unacceptably high.

In another aspect of technology relating to fat is the dietary emphasis on limiting dietary fat intake. For those who eat meats, there is increasing emphasis on low fat content meats in the carcasses of the animals employed in food stocks. Much recent efforts have been devoted to the production of beef, pork, poultry and the like with reduced fat content. Breeding patterns are being manipulated and generic engineering of farm animals is being directed at lowering fat content of the

animals. The techniques of fattening of animals intended for table meat production is highly developed, but is gradually being limited by the emphasis on limiting dietary fats and interest in leaner carcass animals.

Only in very recent times has obesity been addressed in relation to the metabolic pathways of the body and their role and import in fat storage and usage in the body.

Recent research has elucidated some of the mechanisms of obesity and overweight, and has revealed that much of the limitation of prior and current weightloss techniques stems from the fact that they are biochemically and particularly metabolically unsound and incapable of stimulating, regulating and modulating metabolism of fats in adipose tissues. Without these characteristics, it is now known, weight loss and control strategies are likely to fail or to produce conditions as bad as or worse than the weight problems they are intended to alleviate. Without heroic dedication and discipline, and even fanaticism, by the subject, most strategies are short term in their weight loss and control effects.

Increasing efforts have been directed to biochemical research into the mechanisms of fat deposition and metabolism and into stimulating, regulating and modulating metabolism of fats in adipose tissues. Considerable recent progress has been made.

Among the biochemical work of note has been the recent recognition of a role of β -Adreno-receptor activity in the metabolism of fats. It has been recognized that agonists for β -Adrenoreceptors have, in some cases, produced marked weight loss in animals, particularly humans and other mammals.

More recently, the loss of weight has been identified with the β -Adrenoreceptor sub-type, β_3 -Adrenoreceptors. The specific structure of the b3-Adrenoreceptor has been characterized, and demonstrated to be a distinct cellular structure which is Distinguishable from the b1-Adrenoreceptor and the b2-Adrenoreceptor.

It has been demonstrated that compounds which are significant β_3 -Adrenoreceptor agonists produce marked weight loss in animals, and that the loss is sustained with continuation of the administration of such compounds. These

compounds provide potent regulation of fat metabolism. The compounds employed to date are also agonists for the β_1 -Adrenoreceptor and the β_2 -Adrenoreceptor sites. The lack of selectivity represents unwanted side effects of such compounds, and the compounds known as β_3 -Adrenoreceptor agonists to date are not suitable candidates for therapeutic usage because of the unwanted and dangerous side effects.

PROBLEMS AND NEEDS IN THE ART

The existing strategies for weight and body fat regulation are inadequate. The current strategies are ineffective, unsafe, or both. Whether through diet manipulations or through drug usage, or combinations of such strategies, there is a lack of a clear path to safe and effective regulation of body weight and body fat which is safe and effective, which can provide significant and long lasting relief from the health consequences of overweight and obesity and the conditions associated therewith, and from the disease conditions which are aggravated by overweight and obesity.

It is clear that the art lacks and needs therapeutic agents which are highly potent and highly selective β_3 -Adrenoreceptor agonists for effective stimulation, regulation and modulation of metabolism of fats in adipose tissues.

It is also clear that the art lacks and needs agents which are effective β_3 -Adrenoreceptor agonists free of unwanted side effects, and which are safe for stimulating, regulating and modulating metabolism of fats in adipose tissues.

It is clear that the art lacks and needs agents which are effective at regulating the body fat of animals, particularly humans and other mammals, both in the reduction of body weight in the obese and the attendant health problems and issues, and in the production of low fat table meats from domesticated animals for human consumption.

OBJECTS OF THE INVENTION

It is an object of the present invention to provide novel compounds which are safe and effective β_3 -Adrenoreceptor agonists.

It is another object of the present invention to provide syntheses of such β_3 -Adrenoreceptor agonists.

Another object of the present invention is the provision of safe and effective β_3 -Adrenoreceptor formulations for administration to stimulate, regulate and modulate metabolism of fats in adipose tissues in animals, particularly humans and other mammals.

Still another object of the present invention is to provide safe and effective administration of β_3 -Adrenoreceptor agonists for stimulating, regulating and modulating metabolism of fats in adipose tissues in animals, particularly humans and other mammals.

Yet another object of the present invention is to provide a safe and effective regimen for causing and promoting weight loss in humans, and for the maintenance of healthy and personally desired body fat levels.

Still another object of the present invention is to provide safe and effective adjuncts to the husbandry of domesticated animals for the production of low fat dietary meats for human consumption.

The primary objective of the present invention is to provide for weight and body fat regulation through modalities which are effective and safe. The present invention provides a clear path to safe and effective regulation of body weight and body fat which is safe and effective, which can provide significant and long lasting relief from the health consequences of overweight and obesity and the conditions associated therewith, and from the disease conditions which are aggravated by overweight and obesity.

These and related objectives are met by the terms of the present invention as set out in detail in the following specification and defined in the claims appended hereto.

SUMMARY OF THE INVENTION

Compounds which are highly potent and highly specific β_3 -Adrenoreceptor agonists are provided. The compounds are formulated into pharmaceutical

preparations and administered for stimulating, regulating and modulating metabolism of fats in adipose tissues in animals, particularly humans and other mammals.

The compounds of the invention have one of the structures:

$$\begin{array}{c} R_{5} \\ R_{4} \\ R_{7} \\ R_{7} \\ \end{array}$$

$$\begin{array}{c} R_{9} \\ R_{10} \\ \end{array}$$

$$\begin{array}{c} R_{9} \\ R_{10} \\ \end{array}$$

$$\begin{array}{c} R_{1} \\ R_{2} \\ \end{array}$$

Structure A

Structure B

wherein:

- R₁ and R₃ are independently members selected from the group consisting of H, F, CI, Br, I, OCH₃, CF₃, CH₃, alkyl and aryl alkyl;
- R_2 is a member selected from the group consisting of H, I, OCH₃, NH₂, NHR₁₃, NHCOR₁₃, NHCONHR₁₃ and NHCOSR₁₃, and provided that, when both R₁ and R₃ are CF₃, R₂ is not H;
- R_4 and R_5 are each members independently selected from the group consisting of H, OH, F, CI, Br and I, and provided that, when R_4 and R_5 are both OH or both OCH₃, then R_2 is neither NH₂ nor OCH₃;
- $R_{\rm 6}$ and $R_{\rm 7}$ are independently members selected from the group consisting of H, F, CI, Br and I;
- R_8 and R_{13} are independently members selected from the group consisting of H, lower alkyl and aryl alkyl of from 1 to about 8 carbons, F, Cl, Br, I, OCH₃, and CF₃, and provided that, when R_{13} is CH₃, only one or none of R_1 and R_3 is I;
- wherein R₉ and R₁₀ are independently members selected from the group consisting of H, lower straight chain and branched alkyl of from 1 to 8 carbon atoms,

and where R_{11} and R_{12} are independently members selected from the group

consisting of H, lower straight chain and branched alkyl of from 1 to 8 carbon atoms,

and wherein R_1 and R_2 ,taken together, R_2 and R_3 , taken together and R_4 and R_5 , taken together may additionally form a member selected from the group consisting of moieties having the structure:

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wherein R₁₃ and R₁₄ are independently members selected from the group consisting of H, lower straight chain and branched alkyl of from 1 to 8 carbon atoms,

and where R_{11} and R_{12} are independently members selected from the group consisting of H, lower straight chain and branched alkyl of from 1 to 8 carbon atoms,

and the simple inorganic and lower alkyl, of from 1 to about 8 carbons, carboxyllic acid salts thereof.

The preferred compounds of the invention are those wherein one of R_4 and R_5 is OH and the other is H. More preferably, R_5 is OH and R_4 is H. Most preferably, the compound has the following structure:

The invention is also directed to a method and pharmaceutical composition for stimulating, regulating and modulating metabolism of fats in adipose tissues in animals comprising preparing and administering an effective amount of a β_3 -Adrenoreceptor selective agonist which is a member of the group consisting of:

wherein:

- R₁ and R₃ are independently members selected from the group consisting of H, F, CI, Br, I, OCH₃, CF₃, CH₃, alkyl and aryl alkyl;
- R₂ is a member selected from the group consisting of H, I, OCH₃, NH₂, NHR₁₃, NHCONHR₁₃ and NHCOSR₁₃;
- R_4 and R_5 are each members independently selected from the group consisting of H, OH, F, CI, Br and I;
- $\rm R_6$ and $\rm R_7$ are independently members selected from the group consisting of H, F, CI, Br and I;

 R_8 and R_{13} are independently members selected from the group consisting of H, lower alkyl and aryl alkyl of from 1 to about 8 carbons, F, Cl, Br, I, OCH₃, and CF₃;

wherein R_9 and R_{10} are independently members selected from the group consisting of H, lower straight chain and branched alkyl of from 1 to 8 carbon atoms,

and where R_{11} and R_{12} are independently members selected from the group consisting of H, lower straight chain and branched alkyl of from 1 to 8 carbon atoms,

and wherein R₁ and R₂,taken together, R₂ and R₃, taken together and R₄ and R₅, taken together may additionally form a member selected from the group consisting of moieties having the structure:

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wherein R₁₃ and R₁₄ are independently members selected from the group consisting of H, lower straight chain and branched alkyl of from 1 to 8 carbon atoms,

and where R_{11} and R_{12} are independently members selected from the group consisting of H, lower straight chain and branched alkyl of from 1 to 8 carbon atoms,

and the simple inorganic and lower alkyl, of from 1 to about 8 carbons, carboxyllic acid salts thereof.

Preferably, the method and composition use an agonist wherein one of R_4 and R_5 is OH and the other is H. More preferably, R_5 is OH and R_4 is H. Most preferably, the agonist has the following structure:

These compounds are formulated into pharmaceutical carriers to serve as highly selective, effective and safe β_3 -Adrenoreceptor agonists to provide long term weight control.

In humans, the compositions are administered to control body fat levels, and to maintain acceptable body fat levels over time.

In domesticated animals, the compositions are administered to attain desirably low fat content in carcass meats intended for human consumption.

DETAILED DESCRIPTION

The following is a description of the invention, the compounds of the present invention, the method of their synthesis, their formulation into pharmaceutical compositions suitable for administration, and the method of their use for stimulating, regulating and modulating metabolism of fats in adipose tissues in animals, particularly humans and other mammals.

The discussion and presentation of bioactivity information and data in the present description is made in compliance with the standards of the Journal of Medicinal Chemistry. All chemical compounds are named in accordance with the standards of the American Chemical Society rules of standard nomenclature, employing accepted "trivial names" where applicable. All chemical structures are. shown in "skeletal" form, for clarity in understanding the most significant considerations and information about the structures, with implicit hydrogen atoms

not relevant to the conformation of structures not shown, in the fashion typically employed in the Journal of Medicinal Chemistry and most other journals of chemistry. The use of such structural notation is most convenient to understand the structures of such molecules, and those of ordinary levels of skill in the relevant arts are accustomed to such representations and are readily able to identify and understand such "skeletal" structures, including the implicit hydrogen atoms not shown.

INTRODUCTION

The risks and unacceptable levels of adverse consequences of many weight control and weight loss strategies available to individuals and to the medical community make the development of safe and effective modalities for stimulating, regulating and modulating metabolism of fats in adipose tissues an important need in the art and in society as a whole.

The importance of regulating dietary fat intake, and particularly saturated animal fat, has long been recognized. Consumption of meats is primary in the diet in most developed countries, and substantial efforts have been devoted to the development of leaner animals, among other strategies, to facilitate regulating and limiting of dietary intake of saturated animal fats.

In the present invention, the highly desirable goals of stimulating, regulating and modulating metabolism of fats in adipose tissues in animals, particularly humans and other mammals through the modality of administering a pharmaceutical formulation of one or more compounds which are β_3 -Adrenoreceptor selective agonists is provided.

The regulatory and modulatory effect of the compounds of the present invention are dependent on continued administration over time, and the attainment of an equilibrium state which is believed to be dose dependent. In that fashion, the present invention affords the control of body fat in animals, particularly humans and other mammals, over sustained periods, at desirable levels of body fat and/or body mass indices, as defined in the medical literature.

OVERVIEW OF THE INVENTION

Safe and effective control of body fat and body mass indices have been a long sought but quite elusive goal for the medical community. The modalities in use over the past half century have proved to be both dangerous and limited in effectiveness. The longer the effort is sustained, in general, the higher the risk and the lower the effectiveness.

The weight loss effect of β -Adrenoreceptor agonists generally has been known *per se* for a considerable period. That recognition has not led to safe and effective weight loss or regulation because of the copious and highly dangerous side effects.

The recent discovery of the β_3 -Adrenoreceptor and its focal role in fat metabolism holds the promise of the employment of β_3 -Adrenoreceptor agonists in weight loss and regulation. Through the development of compounds which are highly selective for the β_3 -Adrenoreceptor without activation of the β_1 Adrenoreceptor and β_2 Adrenoreceptor the present invention makes that goal attainable.

The β_3 -Adrenoreceptor has not been characterized to date, which makes the search for safe and effective agonists with the required high selectivity a difficult and arduous task. Without a clear understanding of the receptor binding site, the design of effective compounds is based largely on structural activity correlations which are uncertain, unpredictable and unreliable. Even the most minor changes in structure can produce wide deviations in binding affinity, binding specificity, and agonist activity. The compounds of the present invention attain the high affinity for the β_3 -Adrenoreceptor, the low affinity for the β_1 Adrenoreceptor and the β_2 Adrenoreceptor required for effective selectivity and freedom from adverse side effects, and high levels of agonist activity to make the compounds effect in their required role in fat metabolism.

THE β-ADRENORECEPTOR FAMILY

 β Adrenoreceptors have long been known and have been studied for their role in response to the catechol amine hormones adrenaline (epinephrine), noradrenaline (norepinephrine) and dopamine.

Adrenaline, to exemplify the biochemical action of these catechol amine hormones, is a primary agonist for these receptors in the body, and activates metabolic processes within the cells to which it binds. Adrenaline is associated with specific cellular processes which are dependent upon the nature of the cell to which it is bound. The action of adrenaline on the cell is to activate an enzyme within the cell, adenylate cyclase. The adenylate cyclase in turn catalyses further reactions within the target cell, typically beginning an enzyme cascade until the enzyme is broken down or deactivated by cellular regulatory mechanisms. The primary action of adenylate cyclase is the conversion of ATP to cAMP (cyclic adenosine monophosphate or "cyclic adenylate").

In the liver cells, the cAMP activates, in turn, an enzyme cascade which catalyses the conversion of glycogen into glucose and inhibits the conversion of glucose into glycogen, greatly increasing extra-cellular levels of blood glucose in the body.

In muscle tissues, cAMP triggers the breakdown of glycogen into lactate and ATP, providing high levels of ATP to support high levels of muscular activity. In the heart muscle, in particular, the effect is hypertensive and is accompanied by vasodilation throughout the body, increasing blood flow and transport of blood glucose to the cells.

 β -blockers are among the commonly prescribed drugs in the field of cardiology. For the hypertensive patient, competitive binding of the blocking agent to the β Adrenoreceptors modulates and limits the additional hypertensive action of

adrenaline on the heart muscle. The β -blockers may be employed in combination with vasodilators, decreasing the resistance to blood flow peripherally without increasing the heart rate and strength of contraction. A reduction in blood pressure and the work requirement on the heart muscle results.

In the lung, cAMP acts to cause bronchodilation which, when combined with increased blood flow, supplies higher levels of oxygen transport.

(Adrenaline, or epinephrine, is widely employed to stimulate bronchodilation in the treatment of asthma and allergenic reactions which constrict the bronchia.)

Others of the catechol amine hormones have comparable activities.

The release of free fatty acids from adipose tissue has been observed as an action provided by $\boldsymbol{\beta}$ Adrenoreceptor agonists.

A variety of β Adrenoreceptor agonists and blockers have been known for some time, and have proved to be a fruitful field for drug development.

It has been recognized that there are sub-types of the β Adrenoreceptor, designate the β_1 Adrenoreceptor and the β_2 Adrenoreceptor. Lands, et al., "Differentiation of Receptor Systems Activated by Sympathomimetic Amines" Nature, 214:597-598 (1967). Lands, et al., associate the release of free fatty acids from adipose tissue with β_1 Adrenoreceptor activation.

Subsequent studies have provided a spectrum of β Adrenoreceptor agonists and blockers. Among the blockers are both competitive and non-competitive (non-equilibrium) binding agents. Some of such agents are ubiquitous in their action, while others exhibit varying degrees of selectivity for the two sub-types (and hence in the action response produced).

Selective agonist studies show both qualitative and quantitative differentiation of the sub-types. β_1 Adrenoreceptor activation have been demonstrated to cause cardiac stimulation, release of free fatty acids from adipose tissue, and intestinal inhibition. In contrast, β_2 Adrenoreceptor activation produces broncho- and vaso-dilation.

THE β₃-ADRENORECEPTOR

Quite recently, a third sub-type of the β Adrenoreceptor family has been identified. Howe, R. "Beta-3 adrenergic agonists." Drugs Future 1993, 18, 529-549. It has been designated the β 3 Adrenoreceptor. It has also been specifically identified with the release of free fatty acids from adipose tissue, previously attributed by Lands et al. with the β 1 Adrenoreceptor.

While β_1 Adrenoreceptor and β_2 Adrenoreceptor sites are ubiquitous, it has been found that the β_3 -Adrenoreceptor sites are more specialized and are predominantly located on adipose tissue cells, and from studies to date appear to be rather specifically associated with the metabolism of fats.

β₃-ADRENORECEPTOR AGONISTS

This discovery leads quite directly to the search for selective and potent agonists for the β_3 Adrenoreceptor for the treatment of obesity and control of weight. The search is hindered by the lack of characterization of the receptor, but the information from binding studies and other work on β Adrenoreceptor agonists generally indicates that β_3 Adrenoreceptor agonists should be similar in structure to the catechol amine hormones.

Rather little has been published to date on β_3 Adrenoreceptor agonists. See, however, Howe, R. "Beta-3 adrenergic agonists" *Drugs Future* 1993, *18*, 529-549. It is accordingly necessary to extrapolate from the information available about β_1 Adrenoreceptor and β_2 Adrenoreceptor agonists, and to engage in an attempt to discern structural and activity relationships from the available data. The following comments on β_1 Adrenoreceptor and β_2 Adrenoreceptor considerations summarizes what is known in the literature upon which the effort to develop β_3 -Adrenoreceptor agonists can be based.

Trimetoquinol is a potent nonspecific β -adrenoreceptor (β -AR) agonist clinically used in Japan as a bronchorelaxant. Iwasawa, Y.; Kiyomoto, A. "Studies of tetrahydroisoquinolines (THI) 1. Bronchodilator activity and structure-activity relationships." *Jap. J. Pharmacol.* 1967, 17, 143-152. Optical resolution of trimetoquinol and subsequent evaluation of the stereoisomers revealed that the (S)-(-)-isomer of trimetoquinol is a potent β -adrenoreceptor agonist in heart and lung

tissues; whereas, the (R)-(+)-isomer acts as a selective and highly stereospecific thromboxane A₂/prostaglandin H₂ (TP) receptor antagonist. Yamamoto, E.; Hirakura, M.; Sugasawa, S. "Synthesis of 6,7-dihydroxy-1,2,3,4tetrahydroisoquinoline derivatives" Tetraheron Suppl. 1966, 8 (Part 1), 129-134. Mayo, J. R.; Navaran, S. S.; Akbar, H.; Miller, D. D.; Feller, D. R. "Stereodependent inhibition of human platelet function by the optical isomers of trimethoquinol" Biochem. Pharmacol. 1981, 30, 2237-2241. Ahn, C. H.; Romstedt, K. J.; Wallace, L. J.; Miller, D. D.; Feller, D. R. "Characterization of the inhibition of U46619-mediated human platelet activation by the trimetoquinol isomers. Evidence for endoperoxide/thromboxane A2 receptor blockade" Biochem Pharmacol 1988, 37, 3023-33. Shin, Y.; Romstedt, K. J.; Miller, D. D.; Feller, D. R. "Stereodependent antagonism of thromboxane A₂/prostaglandin H₂ receptor sites by trimetoquinol isomers in human platelets, rat vascular endothelial cells and rat vascular smooth muscle cells" Pharmacol. Commun. 1993, 1, 303-312. Radioligand competition binding studies at β -adrenoreceptor and TP receptors show high stereoselective binding (>100-fold) for the S(-)-isomer and R(+)-isomer, respectively. This stereoselectivity is also observed in the binding of fluorinated trimetoquinol analogs at β-adrenoreceptor. Clark, M. T.; Adejare, A.; Shams, G.; Feller, D. R.; Miller, D. D. "5-fluoro- and 8-fluorotrimetoquinol: selective beta 2-adrenoceptor agonists" J Med Chem 1987, 30, 86-90.

Trimetoquinol

The basic catechol structure of catecholamine hormones, such as epinephrine, norepinephrine, dopamine, and the β -adrenoreceptor agonist isoproterenol, is incorporated within the tetrahydroisoquinoline nucleus of trimetoquinol. In studies using mutated hamster β_2 Adrenoreceptor expressed in

Chinese hamster ovary (CHO) cells, replacement of Asp113 with Asn113 abolished receptor binding of trimetoquinol and its analogs. Fraundorfer, P. F. "Functional and biochemical characterization of trimetoquinol (TMQ) analog interactions with β -adrenergic receptor subtypes" Ph. D. Thesis, The Ohio State University, 1993 ("Fraundorfer-2"). In addition, replacement of Ser204 and Ser207 with Ala204 and Ala207 decreased the binding affinity of trimetoquinol analogs in β_2 Adrenoreceptor to a lesser extent, but greatly diminished their ability to stimulate cAMP accumulation. "Fraundorfer-2", supra. However, both the binding and functional activities of isoproterenol are significantly reduced in the β_2 adrenoreceptor Asn113, Ala204 and Ala207 mutants. These results suggest that although trimetoquinol analogs may interact with the same amino acid residues in the binding site as isoproterenol, the contribution of catechol interactions with these mutated β_2 Adrenoreceptors is less significant in terms of ligand binding and may well be overshadowed by the binding contributions of the trimethoxybenzyl group of trimetoquinol.

Substitution with fluorine or iodine on the 5- or 8-positions of trimetoquinol resulted in only a modest (~ 10-fold) increase in β_2 Adrenoreceptor versus β_1 adrenoreceptor selectivity as compared to trimetoquinol in functional and binding studies. Clark, et al., supra; Fraundorfer, P. F.; Fertel, R. H.; Miller, D. D.; Feller, D. R. "Biochemical and pharmacological characterization of high-affinity trimetoquinol analogs on guinea pig and human beta adrenergic receptor subtypes: evidence for partial agonism" J Pharmacol Exp Ther 1994, 270, 665-74.. In addition, it has also found that replacement of the 3'- and 5'-methoxy substituent of trimetoquinol with iodine atoms (i.e., 2) is well tolerated on both β -adrenoceptor, Fraundorfer, et al., supra, and TP receptors. Shin, Y.; Romstedt, K. J.; Miller, D. D.; Feller, D. R. "Interactions of nonprostanoid trimetoquinol analogs with thromboxane A₂/prostaglandin H₂ receptors in human platelets, rat vascular endothelial cells and rat vascular smooth muscle cells" J Pharmacol Exp Ther 1993, 267, 1017-23.; Harrold, M. W.; Gerhardt, M. A.; Romstedt, K.; Feller, D. R.; Miller, D. D. "Synthesis and platelet antiaggregatory activity of trimetoquinol analogs as endoperoxide/thromboxane A2 antagonists" Drug Des Deliv 1987, 1, 193-207.

Interestingly, although its binding affinity at β_1 adrenoreceptor is slightly better than trimetoquinol, compound 2 displays a much higher affinity than trimetoquinol for β_2 adrenoreceptor:

These earlier findings suggest that trimetoquinol analogs interact with an auxiliary site through the substituted benzyl group in addition to the binding site shared by catecholamines. This subsite can be used to advantage in the development of more site-selective agents. The high potency of compound 2 seems to suggest that this auxiliary site is hydrophobic in nature. On TP receptors, the complementary binding sites for trimetoquinol analogs are essentially unknown. However, compound 2 is a more potent TP receptor antagonist than trimetoquinol further suggesting that 1-benzyl ring modifications are appropriate to develop agents with greater selectivity on β -adrenoreceptor versus TP receptors and vice versa.

The literature describes the synthesis and evaluation of iodinated trimetoquinol analogs designed as probes for characterizing the receptor binding interactions, associated with the benzyl substituent of trimetoquinol analogs and as site-selective β-adrenoreceptor and TP receptor ligands. These chemical modifications provide a greater separation of the pharmacological activities for this class of compounds. Site-selective β-adrenoreceptor agents have potential in the treatment of cardiopulmonary diseases, non-insulin dependent diabetes (Type II) and obesity, Howe, R., "Beta-3 adrenergic agonists" *Drugs Future* 1993, *18*, 529-549, whereas highly selective TP receptor antagonists have value in the treatment of thrombolytic disorders. Shin, *supra*; Shin, Y.; Romstedt, K. J.; Miller, D. D.; Feller, D. R., "Interactions of nonprostanoid trimetoquinol analogs with thromboxane A₂/prostaglandin H₂ receptors in human platelets, rat vascular endothelial cells and rat vascular smooth muscle cells" *J Pharmacol Exp Ther* 1993, *267*, 1017-23; Shin, Y.; Romstedt, K.; Doyle, K.; Harrold, M.; Gerhardt, M.; Miller, D.; Feller, D.,

"Pharmacologic antagonism of thromboxane A₂ receptors by trimetoquinol analogs." *Chirality* 1991, 3, 112-117.

Other known β_1 -adrenoreceptor and β_2 -adrenoreceptor agonists include isoproterenol, X and Y, having the structures:

Isoproterenol

While these compounds are highly active β_3 -adrenoreceptor agonists, they are also non-selective, and also bind and activate the β_1 -adrenoreceptor and β_2 -adrenoreceptor with comparable affinities and activities. They are thus entirely unsuited for use in the present invention, but they do afford good basis for comparative and competitive binding studies, and are employed in the present invention for those purposes when appropriate.

THE COMPOUNDS OF THE INVENTION

The present invention is based on the provision of β_3 -adrenoreceptor agonists in pharmaceutically acceptable carrier formulations for administration to stimulate, regulate and modulate metabolism of fats in adipose tissues in animals, particularly humans and other mammals.

The present invention additionally provides a method for safe and effective administration of β_3 -Adrenoreceptor agonists for stimulating, regulating and modulating metabolism of fats in adipose tissues in animals, particularly humans and other mammals.

The present invention provides potent, highly selective β_3 -Adrenoreceptor agonists which are compounds having the structures:

$$R_{5}$$
 R_{4}
 R_{7}
 R_{8}
 R_{10}
 R_{10

R₁ and R₃ are independently members selected from the group consisting of H, F, CI, Br, I, OCH₃, CF₃ and CH₃, alkyl and aryl alkyl;

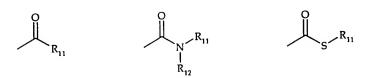
 R_2 is a member selected from the group consisting of H, I, OCH₃, NH₂, NHR₈, NHCOR₁₃, NHCONHR₁₃ and NHCOSR₁₃, and provided that, when both R₁ and R₃ are CF₃, R₂ is not H;

 R_4 and R_5 are each members independently selected from the group consisting of H, OH, F, CI, Br and I, and provided that, when both R_4 and R_5 are OH, then R_2 is neither NH₂ nor OCH₃;

 R_6 and R_7 are independently members selected from the group consisting of H, F, CI, Br and I;

R₈ and R₁₃ are independently members selected from the group consisting ofH, lower alkyl of from 1 to about 8 carbons, F, Cl, Br, I, OCH₃, and CF₃.

wherein R₉ and R₁₀ are independently members selected from the group consisting of H, lower straight chain and branched alkyl of from 1 to 8 carbon atoms,



and where R_{11} and R_{12} are independently members selected from the group consisting of H, lower straight chain and branched alkyl of from 1 to 8 carbon atoms,

and wherein R_1 and R_2 ,taken together, R_2 and R_3 , taken together and R_4 and R_5 , taken together may additionally form a member selected from the group consisting of moieties having the structure:

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COOR_{13}
\end{array}$$

wherein R_{13} and R_{14} are independently members selected from the group consisting of H, lower straight chain and branched alkyl of from 1 to 8 carbon atoms,

and where R_{11} and R_{12} are independently members selected from the group consisting of H, lower straight chain and branched alkyl of from 1 to 8 carbon atoms.

and the simple inorganic and lower alkyl, of from 1 to about 8 carbons, carboxyllic acid salts thereof.

It is preferred that the compounds of the present invention be further qualified and limited to those with high selectivity and high activity for the β_3 -Adrenoreceptor

In addition, there are several particularly preferred species, i.e., specific compounds, which are preferred. These particularly preferred species include the following compounds:

HO.

Formula A 12 Formula A 11 Formula A 10 HO. HO. HO. Formula A 15 Formula A 14 Formula A 13 HO HO. HO НО НО OH Formula A 18 Formula A 17 Formula A 16 HO. Formula A 21 Formula A 20 Formula A 19 Formula A 24 Formula A 23 Formula A 22

Formula A 34

Formula A 35

Formula A 33

Formula A 45

Formula A 46

HOCC N
$$NH_2$$
 NH_2 NH_3 NH_4 NH_2 NH_4 NH_4 NH_5 NH_5

Of these compounds, the following are preferred, A1, A2, A3, A4, A5, A6, A7, A8, A9, A10, A20, A21, A22, A23, A24, A25, A26, A27, A30, A31, A32, A33, A34, A36, A37, A38, A41, A42, A44, A45, A46, and all of the B compounds. At present, the most preferred compound is A4.

Synthesis of TMQ Derivatives

A convenient protection scheme has been devised for the synthesis of the desired β₃-Adrenoreceptor agonists of the present invention. The triple protected isoquinoline intermediates were synthesized as shown in Scheme 1. The tetrahydroisoquinolines 6a-c were synthesized from the O-methyl or O-benzyl protected catecholamines 3a or 3b, respectively, and 4-nitrophenylacetic acid (4a) or 3,5-bis-trifluoromethylphenylacetic acid (4b) using methods described previously. Clark, M. T.; Adejare, A.; Shams, G.; Feller, D. R.; Miller, D. D. "5-fluoro- and 8fluorotrimetoquinol: selective beta 2-adrenoceptor agonists" J Med Chem 1987, 30, 86-90.; Harrold, M. W.; Gerhardt, M. A.; Romstedt, K.; Feller, D. R.; Miller, D. D. "Synthesis and platelet antiaggregatory activity of trimetoquinol analogs as endoperoxide/thromboxane A2 antagonists" Drug Des Deliv 1987, 1, 193-207. Adejare, A.; Miller, D. D.; Fedyna, J. S.; Ahn, C. H.; Feller, D. R. "Syntheses and beta-adrenergic agonist and antiaggregatory properties of N-substituted trimetoquinol analogues" J Med Chem 1986, 29, 1603-9. The amino group of 6a and **6b** were protected with trifluoroacetyl (TFA) and t-butyloxycarbonyl (t-BOC), respectively. The nitro groups of 7a,b were reduced via catalytic hydrogenation using Pd/C or Raney Nickel, respectively, to give the aniline derivatives 8a,b. lodination of 8a,b with 1 equivalent of benzyltrimethylammonium dichloroiodate (BTMACI2I) according to Kajigaeshi et al., Kajigaeshi, S.; Kakinami, H.; Fujisaki, S.;

Okamoto, T. "Halogenation using quaternary ammonium polyhalides. VII. Iodination of aromatic amines by use of benzyltrimethylammonium dichloroiodate (I⁻)" *Bull. Chem. Soc. Jpn.* **1968**, *61*, 600-602, led to the 3'-iodo analogs **9a,b**. An additional 3 equivalents of BTMACl₂I added in portions over a 3 day period was required to convert **8a** completely to the diiodo derivative **10a**.

R action Scheme 1:

Notes to Reaction Scheme 1:

- (a) Toluene, reflux (Dean-Stark trap), 72 hours;
- (b) POCl₃, MeCN, reflux;
- (c) NaBH₄, MeOH;

Example 2

(d) TFAA, THF (Example 3) or (Boc)₂O (NaOH, THF (Example X);

Example 33

- (e) H₂, Pd/C (Example 5) or Raney Ni (Example 6);
- (f) 1 equiv. Of BTMACl₂I, MeOH, CH₂Cl₂, 20 hours (Examples Examples 7, 8), or 4 equiv. Of BTMACl₂I, CaCO3, M⁻OH, CH₂Cl₂, 3days (Example 7)
- (g) 1. BBr₃, CH₂Cl₂, 2. MeOH

Interestingly, the diiodo product 10a, was often isolated as light pink to reddish crystals. It has been found that the minor side product 11 (Scheme 2) was

responsible of the reddish coloration. TLC analysis of the reaction mixture and isolated crude product indicate that compound 11 is formed mostly during work-up. Compound 11 was isolated by flash chromatography. The structure of 11 and its deacetylation product 12 was proven by ¹H and ¹³C NMR and elemental analysis. Compound 11 was also isolated in an attempt to convert the 4'-amino of 10a to a hydrazine group. Thus, diazotization of 10a, followed by reduction with H₂SO₃ gave compound 11 as the only isolated product in low yield.

Reaction Scheme 2:

Notes to Reaction Scheme 2:

- (a) 1. NaNO₂, H₂SO₄, AcOH, 2. H₂SO₃
- (b) NaNO₂, H₂SO₄, H₂O

While reaction of 10a with acetic anhydride at room temperature did not give the desired 4'-acetamido derivative 13, heating 10a in acetic anhydride at reflux resulted in the diacetylation product 16 (Scheme 3). Similar diacetylation has been reported with the reaction of 2,6-dibromo-4-toluidine with refluxing acetic anhydride while lower temperatures gave a mixture of mono and diacetylated products. Ulffers, F.; von Janson, A. Diacetylderivate einger Amine der aromatischen Reihe *Ber.* 1894, 27, 93-101. With this in mind, mono-acetylation was accomplished by reacting 10a with 5 equivalents of acetyl chloride in the presence of 4-dimethylaminopyridine (DMAP) and triethylamine at room temperature to afford 13. Basic hydrolysis of the trifluoroacetyl protecting group of 10a and 13 gave 20c and 14, respectively. The methoxy derivatives 20c, 14, and 6c were demethylated with BBr3 to afford the desired trimetoquinol analogs 21c, 15, and 27, respectively, as hydrobromide salts

(Sch m 1 and 3). In a similar manner, the 6,7-dibenzyloxy-1-(3,5-diiodo-4-methoxybenzyl)-1,2,3,4-tetrahydro-isoquinoline, Harrold, M. W.; Gerhardt, M. A.; Romstedt, K.; Feller, D. R.; Miller, D. D. Synthesis and platelet antiaggregatory activity of trimetoquinol analogs as endoperoxide/thromboxane A2 antagonists *Drug Des Deliv* 1987, 1, 193-207, was dealkylated with BBr3 to give 6,7-dihydroxy-1-(3,5-diiodo-4-hydroxybenzyl)-1,2,3,4-tetrahydroisoquinoline (18) the desmethyl analog of 2. Diazotization of 10a (Scheme 3) followed by reaction of the diazonium salt with H3PO2 or potassium iodide (KI) gave the diiodo and triiodo derivatives, 19a and 19b, respectively. Basic hydrolysis of the trifluoroacetyl group of 19a,b as before gave 20a,b. Demethylation of 20a,b with BBr3 proceeded smoothly to give 21a,b. Compound 9a was acylated with acetic anhydride in refluxing benzene to give 22 which was deprotected in the same manner as 14 to give 23 (Scheme 4).

Reaction Scheme 3:

Notes to Reaction Scheme 3:

- (a) AcCl, Et₃N, DMAP:
- (b) K₂CO₃, MeOH, H₂O;
- (c) 1. BBr₃ CH₂Cl₂, 2. MeOH;
- (d) Ac₂O, reflus;
- (e) NaNO₂, H₂SO₄, AcOH, 2. H₃PO₂ or KI

However, attempts to demethylate 23 with BBr3 failed to give the desired product 26a. Surprisingly, the amide bond of 23 was cleaved to give aniline 24. This indicates the importance of both ortho-iodine atoms as a stenc hindrance toward cleavage of the acetamido group of 14 by BBr3. Trimethylsilyliodide (TMSI) was thus employed as a mild reagent for ether cleavage. However, this agent was too weak to effect demethylation of 23; therefore, the catechol *O*-methyl ether protecting groups were changed to benzyl ethers. Hence, compounds 26a and 26b were prepared from the *O*-benzyl and *N-t*-BOC protected 9b (Schem 4). The

acylated compounds 25a and 25b were deblocked using TMSI. Initially, using the procedure of Lott, R. S.; Chauhan, V. S.; Stammer, C. H. "Trimethylsilyl iodide as a peptide deblocking agent" *J. Chem. Soc. Chem. Comm.* 1979, 495-496, (TMSI, MeCN, 50°C 2h) amide 25a gave the desired amide 26a along with a significant amount of the deacetylation product 24. Ordinarily, amides are stable to TMSI. To optimize the selectivity, the TMSI deprotection reaction was monitored by NMR spectroscopy at room temperature. The *O*-benzyl protecting groups were removed within 6h and no cleavage of the amide bond was observed at this temperature for 20 h. Thus, using the following reaction conditions: 4-6 eq. of TMSI, MeCN, room temperature, 6h, 26a and 26b from 25a and 25b were obtained, respectively.

Reaction Scheme 4

Example 31: $R = CH_3$ Example 32: R = Phenyl

Notes to Reaction Scheme 4:

- (a) AcCl, Et₃N
- (b) K₂CO₃, MeOH, H₂O
- (c) 1. BBr₃, CH₂Cl₂; 2. MeOH,
- (d) Ac₂O, Δ or PhCOCI, Et₃N
- (e) 1. TMSI, MeCN; 2. MeOH

The proton NMR spectra of synthesized compounds were quite complicated, especially the 2-*t*-BOC derivatives which displayed complex splitting patterns reflecting two relatively stable conformations with ratios ranging from 5:2 to 5:4, similar to those observed for N-Ac and N-Me substituted tetrahydroisoquinolines,. Dalton, D. R.; Cava, M. P.; Buck, K. T. "Hindered rotation in 1-benzyl-1,2,3,4-tetrahydro-6,7-dimethoxyisoquinolines" *Tett. Lett.* 1965, 2687-2690; Tomita, M.; Shingu, T.; Fujitani, K.; Furikawa, H. "Studies on the alkaloids of menispermaceous plants. CCXVI. Nuclear magnetic resonance spectra of benzylisoquinoline derivatives. (1). N-Methylcoclaurine type bases" *Chem. Phar. Bull.* 1965, *13*, 921-926. However, the ¹³C NMR spectra of 1-benzyl tetrahydroisoquinolines can be easily used for structure identification because of their relative simplicity.

Assignments of signals (final compounds) were made based on the ¹³C-NMR spectra of salsolinol, Iwasa, K.; Kamigauchi, M.; Takao, N. "Metabolism of

salsalinol by tissue cultures of some Papaveraceae" *Phytochemistry* **1991**, *30*, 2973-2975, on effects of substituents in benzene ring, and off-resonance spectra. For 2-TFA derivatives, the chemical shift of the C-3 atom appears as a quartet (4 J C-F ≈ 3.7 Hz) indicating its close proximity to the CF3 group.

It should be noted that all the compounds produced in the foregoing syntheses are racemic mixtures of the stereoisomers, with a site of asymmetry at C₁₀. It has been observed that the binding specificity and activity of the individual isomers will differ, with the (-) species generally the more specific and active as agonists. Typically, the difference will be relatively modest, and the values of the (+/-) racemic mixtures will be intermediate of the values of the individual isomers and are generally equivalent in their bioactivities to the isolated isomers. It is generally preferred to employ the racemic mixture for reasons of economy and simplicity of the synthesis, but the individual isomers are also a part of the present invention. In suitable cases, the individual isomers can be isolated by stereospecific synthesis or by separatory techniques, both of which are *per se* known to those of ordinary levels of skill in the art.

Little has been published on the biochemical action of β_3 -adrenoreceptor agonists and the behavior of these compounds *in vivo* is not entirely clear. What is clear is that the activity of these agents is dependent upon binding to the β_3 -adrenoreceptor. It is also clear that affinity alone is not the sole consideration, as the compounds vary in their selectivity, some also binding β_1 adrenoreceptor and β_2 adrenoreceptor sites, producing unwanted side effects consistent with a role as agonists or blockers of those sites. They also vary in the degree of agonist activity when bound at the β_3 -adrenoreceptor agonists site. The effect is at least analogous to adrenaline binding to the β_3 -adrenoreceptor and the agonist activity provided by adrenaline, but in the present invention is more selective and substantially free of β_1 adrenoreceptor agonist or blocker activity, β_2 adrenoreceptor agonist or blocker activity, or TP agonist or blocker activity.

The mode of action of these agonists when bound at the β_3 -adrenoreceptor site has not been fully characterized. Although there is no wish to be bound thereby, it is believed that the activation of the cellular mechanisms produced by the agonist

activity is the same as that provided by adrenaline, which has been studied. No indications have been seen which are inconsistent with the adrenaline-like agonist behavior, except that the compounds of the present invention are far more selective for the β_3 -adrenoreceptors and minimally bound to the β_1 adrenoreceptor and β_2 adrenoreceptor sites.

The action of the β_3 -adrenoreceptor agonists of the present invention are also more persistent than the effect of adrenaline. The compounds appear to be less readily broken down in vivo, remain bound to the β_3 -adrenoreceptor sites longer than does adrenaline, and continue to be active for a longer prior of time. It is known that the effects of adrenaline are very rapidly induced by the release of adrenaline into the circulation in response to a stimulus, and are nearly equally rapidly dissipated when the release of adrenaline is slowed to base levels in vivo. While these effects pass within seconds or at most a few minutes, the compounds of the present invention typically persist in their action for an interval of up to about two and often four hours.

As those of ordinary skill in the art will recognize, these features are consistent, in part, with the high binding affinities of the present compounds.

As in the case of adrenaline and other compounds observed to activate β_3 -adrenoreceptors in vivo, the present compounds cause the breakdown of adipose tissue at the cellular level, and increase release of glucose into the circulation and, separately, suppress the conversion of carbohydrates into glucose in the liver. Excess levels of glucose in the blood stream are excreted, primarily in the urine, either per se or as "ketone bodies" produced in the liver. These mechanisms are well studied and characterized and do not form a part of the present invention. It is important in the case of diabetics to be aware of these effects and take them into account in the management of diabetes to avoid the assumption that these conditions are an exacerbation of the diabetes or a failure of the diabetes therapies.

The trimethoxybenzyl portion of trimetoquinol was modified by replacing one or more of the methoxy groups with a variety of halogenated substitutions. The effects of these modifications on the receptor binding affinity of trimetoquinol analogs (Table 1) for human β_2 adrenoreceptor, expressed in CHO cells and human

TP receptors (platelets) were determined by radioligand competition binding assays using [125I]-iodocynopindolol (ICYP) and [3H] SQ 29548 as radioligands, respectively.

Most of the modifications made on the trimethoxybenzyl portion of trimetoquinol resulted in enhancement of β_2 Adrenoreceptor affinity. Previously, it was shown that replacement of the 3' and 5'-methoxy groups of trimetoquinol with iodines [i.e., 1 (pKi = 7.36) 2 (pKi = 8.69)] resulted in a more than 20-fold increase in affinity, Fraundorfer, P. F.; Fertel, R. H.; Miller, D. D.; Feller, D. R. "Biochemical and pharmacological characterization of high-affinity trimetoquinol analogs on guinea pig and human beta adrenergic receptor subtypes: evidence for partial agonism" *J Pharmacol Exp Ther* 1994, 270, 665-74.. In the present study, complete replacement of the 3'-, 4'- and 5'-methoxy groups of trimetoquinol (1) with iodine atoms to give the triiodo analog 21b (pKi = 8.82) enhanced β_2 -adrenoreceptor affinity 29-fold versus trimetoquinol (1) but with respect to 2, the additional iodine substituent at the 4'-position adds little to the binding affinity.

Studies on human β_2 Adrenoreceptor indicate that 4'-position substituents of reflecting varying size and chemical properties are well tolerated. Replacement of the 4'-methoxy of 2 with an amino group [i.e., $2 \varnothing \varnothing 21c$ (pKi = 8.81)] did not significantly alter affinity, while replacement with a 4'-acetamido [i.e., 15 (pKi = 8.06)] reduced affinity only 4-fold. A similar replacement with a hydroxy (i.e., 18, pKi = 7.93) reduced affinity about 5-fold as compared to 2. The receptor binding pocket that interacts with substituents at the 4'-position seems to be sufficiently large to accommodate the 4'-benzamido moiety of 26b (pKi = 8.70). Interestingly, the diiodo analog 21a (pKi = 9.52), which lacks a 4'-substituent, exhibits the most potent affinity with a Ki value in the sub-nanomolar range.

It appears that one *meta*-iodo substituent is sufficient to retain high affinity since removing one of the iodo groups of either **21c** or **15** [i.e., **21c** $\varnothing\varnothing$ **24** (pKi = 8.19) or **15** $\varnothing\varnothing$ **26a** (pKi = 8.11)] resulted in only minor shifts in affinity. To determine the nature (hydrophobic or electronic) of the binding contributions of 3' and 5'-substituents (methoxy and iodo), the *bis*-trifluoromethyl analog **27** was synthesized. While the hydrophobic property (π) of the trifluoromethyl group (π)

0.88) is similar to iodine (π = 1.12), this functional group exerts a much stronger electron withdrawing effect. The binding affinity of the *bis*-trifluoromethyl analog 27 (pKi = 5.36) was five orders of magnitude lower than the dioodo analog 21a. Thus, trifluoromethyl substituents at the 3'- and 5'-positions abolish binding affinity. Since, a trifluoromethyl group is similar in size to an iodine atom, the significantly stronger electron withdrawing property of the trifluoromethyl (σ_p = 0.54 versus σ_p = 0.18 for iodine) is likely responsible for the greatly reduced binding affinity of 27. The electron withdrawing effect of the trifluoromethyl substituents on the π -electron system of the aromatic ring may interfere with its capability to form aromatic interactions with the receptor binding site. These aromatic interactions may be more important for binding than hydrophobic interactions.

Although replacement of the 3' and 5'-methoxy groups of trimetoquinol 1, with iodine atoms (i.e., 2) resulted in a 21-fold increase in β_2 adrenoreceptor affinity, a similar increase in binding affinity was not observed at β_1 adrenoreceptor (Table 2). As a result, the diiodo analog 2 exhibits moderate (ca. 40-fold) selectivity for β_2 adrenoreceptor versus β_1 adrenoreceptor. More importantly, the influence of a 4'-substituent is markedly different for β_2 adrenoreceptor versus β_1 adrenoreceptor. While the absence of a 4'-substituent (i.e., 21a) does not significantly alter β_1 Adrenoreceptor affinity (pKi = 6.74), the same feature increased β_2 Adrenoreceptor affinity. Consequently, analog 21a displays more than 600-fold selectivity for β_2 Adrenoreceptor versus β_1 Adrenoreceptor, and is the most selective trimetoquinol analog yet reported. These results indicate a remarkable difference in the receptor binding site or pocket of β_2 - and β_1 adrenoreceptor that interacts with substituents at the 4'-position of trimetoquinol analogs.

In general, replacement of the 3' and 5'-methoxy groups of trimetoquinol (1, pKi = 6.79) with iodine to give analog 2 (pKi = 7.33) resulted in only a slight increase (3-fold) in affinity. However, replacement of all three methoxy groups of trimetoquinol with iodines to give the triiodo analog 21b (pKi = 4.22) practically abolished binding to TP receptors. In addition, demethylation of the 4'-methoxy substituent of 2 to give 18 (pKi = 4.72) resulted in a similar 380-fold reduction in binding affinity. The very low binding affinity of 18 is in contrast to a recent

observation, Christoff, J. J. "Part 1: Synthesis of arylalkylguanidines as dopamine agonists, Part 2, Section A: Modifications of trimetoquinol and the effects on betaadrenergic and thromboxane A2 receptor system, Section B: Approaches to the asymmetric synthesis of irreversibly binding iodinated derivatives of trimetoquinol." Ph. D. Thesis, The Ohio State University, 1993, where 6,7-dihydroxy-1-(4'-hydroxy-3'-nitrobenzyl)-1,2,3,4-tetrahydroisoquinoline exhibited good binding affinity. By contrast, substitution of the same methoxy group with an amino moiety (i.e. 21c, pKi = 6.73) resulted in only a 3-fold decrease in affinity. Interestingly, removal of the 4-'substituent of 2 or 21c to give 21a (pKi = 6.75) did not affect binding affinity significantly. Acetylation of the 4'-amino group of 21c was also tolerated as 15 (pKi = 6.45) displayed binding affinity similar to 21c. Thus, while a primary amine, acetamide, or a methoxy group is tolerated at the 4'-position, a free hydroxy group or an iodo group is detrimental to binding affinity. Removal of one of the iodines of 21c and 15 to give 24 (pKi = 6.00) and 26a (pKi =5.83), respectively, resulted in 5-fold decrease in binding affinity, suggesting that hydrophobic interactions of 3' or 5'substituents contribute to binding. However, replacement of the 3' and 5'-iodo groups of 21a with similarly hydrophobic trifluoromethyl substituents resulted in drastic reduction in binding affinity. As with β_2 adrenoreceptor, in terms of contribution to overall binding affinity, hydrophobic interactions appear secondary to aromatic interactions.

Synthesis of Thiazolopyridine Derivatives

The preparation of 2-amino-4,5,6,7-tetrahydrothiazolo[5,4-c]pyridines has been well documented in the literature (Thomae, K. 4,5,6,7-Tetrahydrothiazolo[5,4-c]pyridines. Neth. Appl. 6 610 324, 1967; *Chem. Abstr.* 1968, 68, 49593p. Thomae, K. Analgesic Tetrahydrothiazolo[5,4-c]pyridines. Fr. Addn. 94 123, 1969; *Chem. Abstr.* 1970, 72, 100685g. Hantzsch, A.; Traumann, V. Amidothiazole aus Sulfoharnstoff und Halogenisirten Ketonen, resp. Aldehyden, *Berichte* 1888, *21*, 938-941). They are constructed by two approaches: a) Pictet-Spengler reaction – the condensation and subsequent cyclization between an aldehyde and 2-(2'-amino-4'-thiazolyl)ethylamine derivatives; and b) Hantzsch thiazole synthesis - the

condensation and subsequent cyclization between an α -bromopiperidone derivative and thiourea (Figure 3).

Figure 3

However, it was envisioned that both of these approaches would not meet our synthetic challenge. The Pictet-Spengler reaction would necessitate the use of an inherently unstable substituted phenylacetaldehyde whereas, the required α-bromopiperidone for Hantzsch thiazole synthesis is not readily accessible synthetically. Alternatively, the Bischler-Napieralski reaction (Figure 3) is an attractive approach for the synthesis of our designed TMQ analogs, because the required substituted phenylacetic acids are stable, and readily accessible. The Bischler-Napieralski reaction is routinely used to prepare dihydroquinolines (Whaley, W. M.; Govindachari, T. R. The Preparation of 3,4-Dihydroisoquinolines and Related Compounds by the Bischler-Napieralski Reaction, *Org. React.* 1951, *VI*, 74-150).

According to Timmerman's synthetic scheme (Eriks, J. C.; Van der Goot, H.; Sterk, G. J.; Timmerman, H. Histamine H₂ -Receptor Agonists. Synthesis, in Vitro Pharmacology, and Qualitative Sturcture-Activity Relationships of Substituted 4- and 5-(2-Aminoethyl)thiazoles, *J. Med. Chem.* **1992**, *35*, 3239-3246), compound **14**, which is the starting amine for the Bischler-Napieralski approach has been prepared (Scheme 1).

Scheme 1 (a) Methyl vinyl ketone, NaOEt, EtOH, EtOAc, r.t. \rightarrow reflux; (b) (i) Br₂, MeOH, O °C \rightarrow r.t., (ii) 10.0 M H₂SO₄, r.t.; (c) Thiourea, acetone, r.t.; (d) 30% HBr, reflux.

The transformation from **12** to **13** did not proceed in good yield as described. Instead, the procedure described by Sprague et al (Sprague, J. M.; Land, A. H.; Ziegler, C. Derivatives of 2-Amino-4-methylthiazole, *J. Am. Chem. Soc.* **1946**, *68*, 2155-2159) converted **12** to **13** almost quantitatively.

Scheme 2 depicts the synthesis of analogs 7 and 9. Under the Schotten-Baumann condition, substituted phenylacetic acids 15 or 16 (Harrold, M. W.; Gerhardt, M. A.; Romstedt, K.; Feller, D. R.; Miller, D. D. Synthesis and Platelet Antiaggregatory Activity of Trimetoquinol Analogs as Endoperoxide/Thromboxane A₂ Antagonists, *Drug Des. Deliv.* 1987, 1, 193-207) was allowed to react with compound 14. The amide precursors 17 and 18 were obtained and each was treated with POCl₃ in refluxing acetonitrile, and the putative dihydro intermediate was reduced *in situ* with NaBH₄, giving final compounds 7 and 9 which were purified by crystallization or column chromatography on silica gel.

Scheme 2

COOH

$$X$$
 A, b
 A

Scheme 2°: ° (a) Oxalyl chloride, dry benzene, O °C \rightarrow r.t. \rightarrow reflux; (b) **14**, NaOH, CHCl₃, H₂O, r.t.; (c) POCl₃, CH₃CN, reflux; (d) NaBH₄, MeOH, O °C \rightarrow r.t.; (e) 1.0 M HCl in Et₂O.

The acetamido analog 10 was prepared according to Scheme 3. Acetylation of 18 with acetic anhydride gave precursor 19 that was then subjected to POCl₃ and NaBH₄. Compound 10 was isolated as its maleic acid salt.

Scheme 3

Scheme 3^a: ^a (a) Acetic anhydride, dry benzene, dry CH₃CN, reflux; (b) POCI₃, CH₃CN, reflux; (c) NaBH₄, MeOH, O °C \rightarrow r.t.; (d) Maleic acid, CH₃CN.

Analog 11 (Scheme 4) is a derivative of 2-amino-4,5,6,7-tetrahydrothiazolo[4,5-c]pyridine that is an unknown heterocyclic system. Amide precursor 21 has been prepared from compound 20 and 3,4,5-trimethoxyphenylacetic acid under Schotten-Baumann condition.

Scheme 4

Scheme 4 (a) 3,4,5-Trimethoxyphenylacetyl chloride, NaOH, CHCl₃, H₂O, r.t.; (b) (i) POCl₃, CH₃CN, reflux, (ii) NaBH₄, MeOH, O °C \rightarrow r.t.; (c) P₂O₅, celite, dry CHCl₃, dry benzene, reflux; (d) Acetic anhydride, dry benzene, dry CH₃CN, reflux; (e) (i) BH₃•THF, THF, O °C \rightarrow r.t. \rightarrow reflux, (ii) 1.0 M HCl in Et₂O.

Compound **20** was synthesized according to literature scheme (Scheme 5) (Eriks, J. C.; Van der Goot, H.; Sterk, G. J.; Timmerman, H. Histamine H₂ -Receptor Agonists. Synthesis, in Vitro Pharmacology, and Qualitative Sturcture-Activity Relationships of Substituted 4- and 5-(2-Aminoethyl)thiazoles, *J. Med. Chem.* **1992**, 35, 3239-3246).

Scheme 5

Scheme 5^a: ^a (a) 4-Chloro-1-butanol, dry DMF, 70 °C; (b) (i) Oxalyl chloride, dry CH₂Cl₂, dry DMSO, -60 °C \rightarrow 50 °C, (ii) Et₃N, H₂O, -50 °C \rightarrow r.t.; (c) Br₂, dry CCl₄, r.t.; (d) Thiourea, acetone, r.t. (e) 30% HBr, reflux.

However, the conversion from the α-bromoaldehyde **26** to 2-aminothiazole **27** was again achieved effectively by using the procedure of Sprague et al (Sprague, J. M.; Land, A. H.; Ziegler, C. Derivatives of 2-Amino-4-methylthiazole, *J. Am. Chem. Soc.* **1946**, *68*, 2155-2159). Using various conditions (Whaley, W. M.; Govindachari, T. R. The Preparation of 3,4-Dihydroisoquinolines and Related Compounds by the Bischler-Napieralski Reaction, *Org. React.* **1951**, *VI*, 74-150; Kametani, T.; Fukumoto, K.; Fujihara, M. Studies on the Syntheses of Heterocyclic Compounds. Part CDLIV. Abnormal Dienone-Phenol Rearrangement of . Procularine, *J. Chem. Soc. Perkin Trans. I* **1972**, 394-396; Rice, K. C.; Brossi, A.

Expedient Synthesis of Racemic and Optically Active N-Norreticuline and N-Substituted and 6'-Bromo-N-norreticulines, *J. Org. Chem.* **1980**, *45*, 592-601) of Bischler-Napieralski reaction, attempts to cyclize the amide precursor **21** proved to be unsuccessful. However, it has been reduced by BH₃ to give the open chain analog **24** that has also been examined in all three human β -AR subtypes. Similarly, the acetamido amide precursor **22** also failed to give the cyclized product under the reaction condition described.

COMPETITIVE BINDING STUDIES

The competitive binding studies indicate that the compounds of the present invention are reversibly bound and confirms that they are equilibrium competitive binding species. No evidence of covalent bonding or other non-equilibrium binding mechanisms to the receptor sites is seen.

Binding to the β_1 adrenoreceptor, β_2 adrenoreceptor and the TP receptor sites is low, indicating that few side effects associated with these receptors is observed.

Binding to other types of receptors is unlikely.

Biochemical actions other than those associated with β_3 -adrenoreceptor binding and agonist activity are not observed. In particular, the compounds of the present invention exhibit low acute toxicities, low long-term subacute toxicity, no evidence of receptor down regulation or other loss of effectiveness over time, and no indications to date of adverse side reactions or side effects, short term or long term, which would represent contraindications for the indicated use.

The chemical properties and stability of the compounds permits their formulation into substantially any suitable vehicle or carrier consistent with the intended mode of administration. Those of ordinary levels of skill in the pharmaceutical industry are fully capable of devising appropriate carriers for the administration of the compounds consistent with the intended mode of administration, and such carriers are employed in the present invention, but are not per se an inventive part thereof.

The compounds of the present invention can be administered effectively by any route or modality consistent with the administration of water-soluble, polar

pharmaceutical compounds. Administration may be oral, injectable (via im, iv, ip and subcutaneous, injections), or may be implanted for sustained release by employment of known implantable systems for the administration of bioactive agents. Absorption from suppositories is also effective, although generally limited by patient reluctance.

Topical administration is also effective, although provision must be made for a suitable skin transport agent to be associated with the compounds to assure that an effective level of the compounds are administered to provide useful systemic levels.

Oral administration and, in suitable cases, implantable sustained release systems are the preferred mode of administration. Implants are particularly effective for those with acute obesity, for those who are inconsistent and/or non-compliant with scheduled oral dosages, and related circumstances. For general usage, oral administration will ordinarily be preferred.

At the outset of administration, the compounds of the present invention produce marked weight loss in over-weight subjects, often at rates of greater than one pound per day and in acute subjects, at rates greater than two pounds per day. The rate is generally sustained until the level of adipose tissue is markedly reduced, and the loss of weight and fatty tissue slows until, at an equilibrium point which is directly related to dosage levels, the body weight of the subject stabilizes. The dosage level may require adjustment from time to time to attain a suitable equilibrium level, which may be suitably defined by the percentage of body fat, the "body mass index" as that term is generally defined in the medical literature, or by other known factors.

If the compounds of the present invention are withdrawn, unless "semipermanent" metabolic changes are induced as hypothesized below, the benefits of the present invention will be lost over time, and the body weight and body fat will, unregulated, tend to return to its former level.

Substantially all the adverse effects of over-weight and obesity will be eliminated, or at the least be materially reduced, as the levels of fatty tissues and body mass are reduced.

The problems of glucose intolerance, insulin resistance and Type II diabetes (non-insulin dependent diabetes), in particular, will be reduced in magnitude or eliminated. In particular, the adverse and progressive effects of such conditions will generally be arrested, the risks of non-compliance with specific treatments and dietary management will be reduced or eliminated, and the causes of such conditions may, in some cases, be moderated or even reversed.

As body fat is reduced, and body mass is reduced, the load on joint in the body will be reduced, which may produce beneficial palliation of the effects of osteo-arthritis, rheumatoid arthritis and other joint disorders and sources of pain. Those having limited mobility associated with obesity will be better able to exercise and improved mobility can be expected to result.

In most uses, the present invention will produce improved psychological profiles in those concerned with body image and appearance. Such effects are virtually always healthy and beneficial developments.

As the level of body fat declines, and so long as serum levels of glucose are kept low, there will generally be an increase in the metabolism of dietary fat for the production of catabolic energy requirements. A decline in serum cholesterol, low density lipids, total serum lipids and related problems will occur. The antihypercholersteremic effect is a highly desirable consequence of the practice of the present invention. Such effects can be enhanced by a diet which limits carbohydrate intake. Over time, such effects may limit the progress of or even reduce arterial occlusion and the incidence of ischemic and related consequences, including, among others, stroke and myocardial infarction.

The increased reliance on dietary fats instead of carbohydrates, and particularly serum glucose, may become so substantial, in fact, that in some cases intake of dietary fats will need to be increased to support normal catabolic energy requirements.

It is hypothesized that, over a long term, the effect of the administration of the compounds of the present invention will so limit blood glucose and the conversion of blood glucose into fats that the body will undergo a "semi-permanent" metabolic change, with increased (up-regulated) insulin receptors, decreased glucose

intolerance (increased insulin responsiveness and efficiency generally), and reduced (down-regulated) mechanisms for the conversion of glucose and other carbohydrates into adipose fat tissues. Increased reliance on dietary fat as energy sources for catabolic processes would continue with the alteration of glucose responses and the hypocholesterolemic effects would continue. These effects are not expected to occur until a suitable equilibrium state is attained, and will lag substantially behind body mass and fatty tissue reduction, and will arise only as mitosis produces new cells adapted to the equilibrium conditions, with environmentally "adjusted" receptor populations. These effects would be of particular benefit in preventing or treating Type II diabetes and glucose intolerance, in particular. These effects would alter the body mass equilibrium point in a fundamental fashion, and would require reducing the dosage of the compounds of the present invention and may, with suitable dietary modifications and other behavioral modification, permit weaning the subject from reliance on the compounds altogether,.

All indications to date suggest that the β_3 -Adrenoreceptor agonists of the present invention are fully effective modulators of body weight without reliance on adjunctive or combination therapies. When such additional modalities, such as dietary restrictions and/or exercise, are desirable for other reasons, or where the need for weight regulation and glucose modulation at the cellular level is needed in concert with treatment of other disorders, such as diabetes and the like, there is no evidence that the present compounds and the utilization thereof conflict in any way with such additional or adjunctive treatments and therapies.

In particular, the present compounds do not interfere with the administration of insulin or other agents to diabetic patients.

In addition, the present compounds are not inconsistent with special diets employed to regulate glucose intolerance, insulin intolerance and related disorders, including very low carbohydrate diets, very high protein diets, and the like.

Where appetite and eating disorders are factors, the modalities of the action of the present invention do not conflict with the usual and common treatments and therapies employed to control such conditions, including behavior modification, drug

therapies and surgical interventions, although the present invention may in many cases eliminate the need to resort to such higher risk strategies and the problems and consequences thereof may often be regulated by the present invention without resort to such additional strategies. The present invention results in an intrinsic suppression of appetite, tends to trigger satiety and tends to modulate the factors which promote appetite and eating disorders, and is likely, over time, to result in substantial behavior modification as the causes of such disorders are modulated. The use of appetite suppression is generally not indicated or required with the present invention.

In general, when the compounds of the present invention are administered, it is highly desirable to assure a balanced diet and adequate intake of vitamins and minerals. The nature of the action of the compounds of the present invention can accelerate the excretion of both oil soluble and water soluble vitamins and of minerals. In addition, it is advisable to maintain high fluid intake to aid in the excretion of the high levels of glucose to be eliminated via the kidneys.

It is also appropriate in many cases to induce a suitable exercise program to improve muscle tone, strength, mobility and agility in parallel with the practice of the present invention. While many overweight individuals exercise considerably, many more do not. The compounds of the present invention do not increase muscle strength or endurance, although the user will benefit from the reduced body mass and the attendant load the muscles must carry, so that greater endurance and "perceived strength" or "relative strength" may increase and increased activity levels and exercise levels will often be a desirable side benefit of the practice of the invention.

Particularly those with limited joint functionality as an incident of arthritic conditions and other like causes will find substantial palliation and increased joint functionality as adipose tissue and body mass are reduced and the load on the joints is reduced as a consequence.

The loss of body bulk may provide enhanced joint flexibility and increased range of motion in some subjects who have been limited by obesity.

No absolute contraindications or adverse side effects of the compounds of the present invention have been found to date.

Those with impaired liver or kidney function may face adverse consequences from the increased levels of glucose in the circulation and the increased load on the blood glucose processing and excretion. Such individuals may require reduced dosages (and more gradual effects) to limit the loading on the liver and kidneys. The invention has not, to date, been tested with individuals with partially or totally incompetent kidneys, i.e. kidney deficient dialysis patients, but it appears that the present invention is not likely to exacerbate either the conditions which require reliance on dialysis or the risks of dialysis procedures themselves.

These compounds do not significantly bind or activate the β_1 adrenoreceptor or the β_2 adrenoreceptor and thus do not produce the usual responses of agonists or blockers for these sites. There is no indication of significant or measurable binding to α -adrenoreceptors, and no indication of any activity consistent with such agonist or blocking activity. There is accordingly no hypertensive effect, no vasodilator effect, and no bronchodilator effect observed in connection with the administration of the compounds of the present invention. These observations are, of course, consistent with the high level of selectivity of these compounds for the β_3 -Adrenoreceptors.

EXAMPLES

The following specific examples demonstrate and illustrate the synthesis of the compounds of the present invention and intermediates prepared in the course of the synthesis. The following apply to all the examples related to the syntheses of compounds:

Melting points were determined on a Thomas-Hoover capillary melting point apparatus and are uncorrected. Infrared spectra were recorded on a Perkin Elmer System 2000 FT-IR. Proton and carbon-13 magnetic resonance spectra were obtained on a Bruker AX 300 spectrometer (300 and 75 MHz for ¹H and ¹³C, respectively). Chemical shift values are reported as parts per million (δ) relative to tetramethylsilane (TMS). Spectral data are consistent with assigned structures. Elemental analyses were performed by Atlantic Microlab Inc., Norcross, GA, and

found values are within 0.4% of the theoretical values. Routine thin-layer chromatography (TLC) was performed on silica gel GHIF plates (250 m, 2.5 x 10 cm; Analtech Inc., Newark, DE). Flash chromatography was performed on silica gel (Merck, grade 60, 230-400 mesh, 60 Å). Tetrahydrofuran (THF) was dried by distillation from sodium metal, and acetonitrile (MeCN), CHCl3 and methylene chloride (CH2Cl2) were dried by distillation from P2O5. All anhydrous solvents (except anhydrous Et2O and THF) were stored over 3- or 4-Å molecular sieves.

TMQ Derivatives

EXAMPLE 1 N-(3,4-dimethoxyphenethyl)-4-nitrophenylacetamide

A solution of 3,4-dimethoxyphenethylamine (5.0 g, 27.6 mmol) and 4-nitrophenylacetic acid (7.5 g, 41.4 mmol) in toluene (150 mL) was heated at reflux for 72 h in a flask equipped with a Dean-Stark trap under an argon atmosphere. The solvent was evaporated in vacuo and the residue was taken up in CH₂Cl₂ (200 mL). The solution was washed consecutively with H₂O (100 mL), 10% HCl (2 x 100 mL), H₂O (2 x 100 mL), 10% NaHCO₃ (2 x 200 mL), H₂O (2 x 100 mL) and dried over MgSO₄. The solvent was evaporated and the crude solid was recrystallized from EtOAc to give 5.49g (58%) of the product as ivory colored needles: mp 119-121 °C (lit. 22 130-132°C, ethanol-isopropanol); 1 H NMR (CDCl₃) δ 8.16 (d, J = 8.8 Hz, 2H, ArH), 7.37 (d, J = 8.8 Hz, 2H, ArH), 6.73 (d, J = 8.1 Hz, 1H, ArH), 6.65 (d, J = 1.9 Hz, 1H, ArH), 6.60 (dd, J = 8.1 & 1.9 Hz, 1H, ArH), 5.40 (m, 1H, NH), 3.86 (s, 3H, OMe), 3.84 (s, 3H, OMe), 3.59 (s, 2H, CH₂), 3.51 (q, J = 6.9 Hz, 2H, CH₂), 2.73 (t, J = 6.9 Hz, CH₂); IR (KBr) 3320 (NH), 1650 (C=O) cm- 1 ; Anal. (C18H₂0N₂O₅) C, H, N.

N-(3,4-dim thoxyphenethyl)3,5-bis-trifluoromethylphenylacetamide

A solution of 3,4-dimethoxyphenethylamine (2.72 g, 15 mmol) and 3,5-bistrifluoromethylphenylacetic acid (2.72 g, 10 mmol) in toluene (50 mL) was heated at reflux for 80 h in a flask equipped with a Dean-Stark trap. The solvent was evaporated in vacuo and the residue was taken up in CH₂Cl₂. The solution was washed consecutively with 0.1 N HCl (30 mL), H₂O (50 mL), 0.1 N NaOH (30 mL), H₂O (50 mL) and dried over MgSO₄. The solvent was evaporated and the crude solid was recrystallized from toluene to give 3.44g (79%) of the product as white needles: mp 127-128 °C; 1 H NMR (CDCl₃) 8 7.79 (s, 1H, ArH), 7.72 (s, 2H, ArH), 6.75 (d, 1 J = 8.1 Hz, 1H, ArH), 6.67 (d, 1 J = 1.9 Hz, 1H, ArH), 6.61 (dd, 1 J = 8.1 & 1.9 Hz, 1H, ArH), 5.55 (m, 1H, NH), 3.85 (s, 3H, OMe), 3.84 (s, 3H, OMe), 3.58 (s, 2H, CH₂), 3.52 (q, 1 J = 6.9 Hz, 2H, CH₂N), 2.75 (t, 1 J = 6.9 Hz, CH₂); 1 3C NMR (CDCl₃) 8 168.72, 149.20, 147.87, 137.28, 131.90, 130.79, 129.47, 123.16, 121.20, 120.56, 111.72, 111.27, 55.83, 42.85, 40.90, 34.97; IR (KBr) 3323 (NH), 1651 (C=O) cm⁻¹; Anal. (C₂OH₁₃F₆NO₃) C, H, N.

EXAMPLE 3

6,7-Dimethoxy-1-(4-nitrobenzyl)-1,2,3,4-tetrahydroisoquinoline

A mixture of 5a (8.0g, 23.2 mmol) and POCl3 (15.6 mL, 167.4 mmol) in dry MeCN (160 mL) was heated at reflux for 4 hours. The solvent was evaporated in

vacuo to give a glassy residue which was taken up in methanol (250 mL) and evaporated to dryness three times until the residue was a solid. The solid residue was dissolved in MeOH (250 mL) then cooled in an ice bath. Excess NaBH4 (17.56g, 167.4 mmol) was carefully added in portions. The mixture was stirred at room temperature overnight. The solvent was removed in vacuo and the solid residue was partitioned in CH2Cl2 (250 mL) and H2O (150 mL). The layers were separated and the H2O layer was extracted with CH2Cl2 (100 mL). The combined organic fraction was washed successively with H2O (2x50 mL), 2N NaOH (2x50 mL), H2O (50 mL), and dried with Na₂SO₄. The solvent was evaporated to give a reddish oil. The oil was taken up in a minimum amount of methanol. The product crystallized upon standing and was collected by filtration (3.02g, 40%): mp 134-36 °C; ¹H NMR (CDCl₃) δ 8.18 (d, J=8.7 Hz, 2H, ArH), 7.43 (d, J=8.7 Hz, 2H, ArH), 6.62 (s, 1H, ArH), 6.61 (s, 1H, ArH), 4.44 (dd, J=9.5, 4.1 Hz, ArCH-N), 3.87 (s, 3H, OMe), 3.84 (s, 3H, OMe), 3.28 (dd, J=13.7, 4.1 Hz, 1H, ArCH₂), 3.23-3.15 (m, 1H, NCH), 3.04 (dd, J=13.7, 9.5 Hz, 1H, ArCH), 3.00-2.91 (m, 1H, NCH), 2.71 (m, 2H, ArCH₂); Anal. (C₁₈H₂₀N₂O₄) C, H, N.

EXAMPLE 4

6,7-Dimethoxy-1-(3,5-bis-trifluoromethybenzyl)-1,2,3,4-tetrahydroisoquinoline hydrochloride

The amide 5c (1.31 g, 3 mmol) was cyclized in the same manner as 6a (7 mL of 1M HCl in ether was added to the methanolic solution of a crude product) to give 6c (0.84 g, 60%) as a hydrochloride salt: mp 104-115 °C (MeOH-ether); 1 H NMR (CDCl3) 8 10.34 (bs, 2H, NH), 7.82 (s, 1H, ArH), 7.75 (s, 2H, ArH), 6.61 (s, 1H, ArH), 5.87 (s, 1H, ArH), 4.77 (m, 1H, CH), 3.91 (m, 1H, CH), 3.84 (s, 3H, OMe), 3.47 (s, 3H, OMe), 3.44 (m, 2H, CH), 3.28 (m, 2H, CH), 3.02 (m, 1H, CH); 13 C NMR

(CDCl₃) δ 149.29, 147.74, 138.74, 132.08, 130.39, 123.33, 123.07, 121.40, 111.71, 109.43, 55.92, 55.38, 54.94, 40.46, 38.30, 24.80; IR (KBr) 3436 (NH), 1281, 1379 (C-O) cm⁻¹. Anal. (C₂₀H₁₉F₆NO₂•HCl•0.5 H₂O) C, H, N.

EXAMPLE 5

6,7-Dimethoxy-1-(4-nitrobenzyl)-2-trifluoroacetyl-1,2,3,4- tetrahydroisoquinoline

A solution of 6,7-dimethoxy-1-(4-nitrobenzyI)-1,2,3,4-tetrahydrisoquinoline (6a) (3.0g, 9.14 mmol) in dry THF (150 mL) was added to trifluoroacetic anhydride (20 mL) with stirring at 0°C. The mixture was stirred at room temperature overnight with the flask equipped with a CaCi₂ drying tube. The reaction mixture was poured onto ice (200g) and the mixture stirred for 30 minutes. CH₂Cl₂ (200 mL) was added and stirring was continued for 10 minutes. The layers were separated and the organic layer was washed consecutively with H₂O (50 mL), 0.2N NaOH (100 mL), H₂O, (100 mL), and then dried with Na₂SO4. The solvent was evaporated in vacuo to give a yellow solid. Recrystallization from EtOAc-MeOH gave 1.94g (50%) of yellow crystals: mp 162-64 °C; ¹H NMR (CDCl₃) δ 8.14 (m, 2H, ArH), 7.28 (m, 2H, ArH), 6.62 (s, 1H, ArH), 6.34 (s, 1H, ArH), 5.64 (t, *J* =6.7 Hz, ArCH-N), 3.87 (s, 3H, OMe), 3.72 (s, 3H, OMe), 3.3-3.6 (m, 2H, N-CH₂), 3.25 (d, 2H, ArCH₂), 2.98-2.6 (m, 2H, ArCH₂); IR (KBr) 1686 (C=O), 1519, 1340 (NO₂) cm-¹; MS m/e (m+): 423 (M+H, FAB). Anal. (C₂0H₁9F₃N₂O₅) C, H, N.

6,7-Dibenzyloxy-2-tert-butoxycarbonyl-1-(4-nitrob nzyl)-1,2,3,4-tetrahydroisoquinoline

A solution of (Boc)₂O (2.84 g, 13 mmol) in THF (10 mL) was added to a cold mixture (ice bath) of isoquinoline **6b** (6.20 g, 12 mmol) in THF (100 mL) and 1N NaOH solution (30 mL). The ice bath was removed and stirring was continued at room temperature overnight. THF was evaporated under reduced pressure, water was added and the product was extracted with CH₂Cl₂, dried over MgSO₄, filtered and evaporated again. The oily residue was dissolved in ether and put in a refrigerator. Pink crystals were filtered and washed with ether to give 6.00 g (86 %) of the title compound: mp 150-152 °C; ¹H NMR (CDCl₃) δ (the spectrum consists of two rotamers of 5:4 ratio) 8.11 and 8.06 (d, J = 8.2 Hz, 2H, ArH), 7.47-7.27 and 7.21-7.11 (m, 12H, ArH), 6.70 and 6.67 (s, 1H, H-5), 6.56 and 6.44 (s, 1H, ArH), 5.27-4.96 (m, 5H, CH₂O + CH), 4.12 and 3.74 (m, 1H, CH), 3.25-3.00 (m, 3H, CH, CH₂Ar), 2.87-2-60 (m, 1H, CH), 2.57-2.37 (m, 1H, CH), 1.38 and 1.25 (s, 9 H, t-Bu); IR (KBr) 1688 (C=O), 1518, 1345 (NO₂) cm⁻¹. Anal. (C35H36N₂O₆) C, H, N.

EXAMPLE 7

1-(4-Aminobenzyl)-6,7-dimethoxy-2-trifluoroacetyl-1,2,3,4- tetrahydroisoquinoline

A solution of **7a** (5.20g, 12.24 mmol) in ethyl acetate (200 mL) was hydrogenated (60 psi) over 5% Pd/C (1g) for 2 hours. The catalyst was removed by filtration and the filtrate was evaporated to dryness to give a beige solid. Recrystallization from ethyl acetate and hexane gave 4.20 (87%) of the product as light pink to white crystals: mp 157-160 °C; ¹H NMR (CDCl3) δ 6.88 (d, 2H, ArH), 6.59(d, 3H, ArH), 6.32 (s, 1H, ArH), 5.53 (t, 1H, ArCH-N), 3.99 (m, 1H, CH), 3.86 (s, 3H, OMe), 3.71 (s, 3H, OMe), 3.60 (bs, 2H, NH₂), 3.42-3.56 (m,2H,CH), 2.85-3.20 (m, 3H,CH), 2.59-2.73 (m, 1H, CH); IR (KBr) 3370 (m, NH₂), 1689 (C=O) cm⁻¹; MS m/e (m+): 395 (M+H, FAB). Anal. (C₂₀H₂₁F₃N₂O₃)C, H, N.

EXAMPLE 8

1-(4-Aminobenzyl)-6,7-dibenzyloxy-2-tert-butoxycarbonyl-1,2,3,4-tetrahydroisoquinoline

The nitro compound **7b** (6.00 g, 10.3 mmol) was dissolved in EtOAc (230 mL) in a Parr bottle. The solution was charged with a slurry of Raney-Ni (4 mL) and hydrogenated at 50 psi for 3h. The solution was filtered through celite and evaporated to give 5.10 g (90 %) of the crude compound. The product was purified by flash chromatography (silica gel, hexane-EtOAc 2:1) to give a foamy glassy solid (4.51 g, 71 %); ¹H NMR (CDCl₃) δ (the spectrum consists of 2 rotamers of 5:2 ratio) 7.48-7.24 (m, 10H, 2xPh), 6.82 (m, J = 8.2 Hz, ArH, 6.68 and 6.64 (s, 1H, ArH), 6.58 (m, J = 8.2 Hz, 2H, ArH), 6.49 and 6.32 (s, 1H, ArH), 5.12-4.81 (m, 5H, 2xCH₂O + CH), 4.18-4.08 and 3.81-3.71 (m, 1H, CH), 3.27-3.09 (m, 1H, CH), 3.00-2.60 (m, 3H, CH₂Ar, CH), 2.59-2.40 (m, 1H, CH), 1.43 and 1.32 (s, 9H, t-Bu); IR (KBr) 3451 and 3365 (NH₂), 1684 (C=O), 1624 (NH bend), 1517 (C=C Ar) cm⁻¹. Anal. (C35H38N₂O₄) C, H, N.

1-(4-Amino-3-iodobenzyl)-6,7-dibenzyloxy-2-tert-butoxycarbonyl-1,2,3,4-tetrahydroisoquinoline

A mixture of isoquinoline 8b (1.11 g, 3.2 mmol), BTMACl2I (1.1g, 3.2 mmol), CaCO3 (0.44 g, 4.4 mmol) in CH2Cl2 (50 mL) and MeOH (20 mL) was stirred overnight at room temperature. CaCO3 was filtered and washed with CH2Cl2. The filtrate was washed with solution of Na₂S₂O₃ (x 2), H₂O (x 2), dissolved in CHCl₃ and EtOH and concentrated till the beginning of crystallization to give 1.59 g (81%) of title compound as pink crystals: mp 169-171 °C; 1 H NMR (CDCl₃) 8 (the spectrum consists of 2 rotamers of 2:1 ratio) 7.47-7.25 (m, 11H, ArH), 6.81 (dd, 1 J = 8.1, 1 J = 1.6 Hz, 1H, ArH), 6.70-6.59 (m, 2H, ArH), 6.49 and 6.30 (s, 1H, ArH), 5.20-5.85 (m, 5H, CH₂O + CH), 4.13 and 3.74 (m, 1H, CH), 4.01 (s, NH₂), 3.30-3.10 (m, 1H, CH), 2.96-2.59 (m, 3H, CH₂Ar + CH), 2.59-2.43 (m, 1H, CH), 1.44 and 1.32 (s, 9H, t-Bu); IR (KBr) 3453 and 3334 (NH₂), 1667 (C=O), 1627 (NH bend), 1520 and 1498 (C=C Ar) cm⁻¹. Anal. (C₃5H₃7IN₂O₄) C, H, N.

1-(4-Amino-3,5-diiodobenzyl)-2-trifluoroacetyl-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline

To a solution of 8a (0.1 g, 2.54 mmoL) in CH₂Cl₂ (50 mL) and methanol (20 mL)was added BTMACl₂I (2.0 g, 5.77 mmoL) and CaCO₃ (2.0 g). The mixture was stirred overnight at room temperature. A second portion of BTMACl₂I (0.97g, 2.8 mmol) was added and stirring was continued overnight. Analysis of the reaction indicated a mixture of mono- and diiodinated products. The reaction mixture was filtered. The filtrate was washed consecutively with aqueous 5% Na₂S₂O₃ (40 mL) and water (50 mL), then dried with Na₂SO₄. Evaporation of the solvent gave a reddish glassy solid. The desired diiodinated product was purified from the crude mixture by flash chromatography (CH₂Cl₂:EtOAc, 9:1). The appropriate fractions were combined and evaporated in vacuo to give 0.72 (44%) of the product as a white solid: mp 183-184.5 °C; 1H NMR (CDCl₃) δ 7.37 (s, 2H, ArH), 6.61 (s, 1H, ArH), 6.29 (s, 1H, ArH), 5.43 (m, 1H, ArH), 4.56 (bs, 2H, NH₂), 3.87 (s, 3H, OMe), 3.73 (s, 3H, OMe), 3.60 (m, 1H, CH), 2.91 (m, 4H, CH), 2.70 (m, 1H, CH); IR (KBr) 3429, 3348 (NH), 1685 (C=O) cm⁻¹; Anal. (C₂OH₁9F₃I₂N₂O₃) C, H, N.

4',4"-Azobis[1-(4-Amino-3,5-diiodobenzyl)-6,7-dimethoxy-2-trifluoroacetyl-1,2,3,4-tetrahydroisoquinoline]

$$CH_3O$$
 CH_3O
 CH_3

Was isolated from *the above mixture* as a bottom spot, deep-purple solid: mp 229-232°C; 1H NMR (CDCl₃) δ 7.78 (s, 2H, ArH), 6.65 (s, 1H, ArH), 6.18 (s, 1H, ArH), 5.53 (dd, J = 7.9, 5.6 Hz, 1H, ArH), 3.99 (m, 1H, CH), 3.88 (s, 3H, OMe), 3.74 (m, 1H, CH), 3.72 (s, 3H, OMe), 3.16 (dd, J = 13.0, 5.3 Hz, 1H, CH), 2.96 (m, 2H, CH), 2.80 (dt, J = 16.2, 4.5 Hz, 1H, CH); ¹³C NMR (CDCl₃) δ 156.14, (q) 148.89, 148.54, 147.59, 142.27, 141.53, 125.60, 125.05, 116.47 (q), 111.28, 110.42, 89.83, 56.15, 56.01, 55.54, 40.88 (q), 40.60, 28.45; IR (KBr) 1688 (C=O), 1520 (C=C Ar) cm⁻¹; Anal. (C40H34F6l4N4O6) C, H, N.

The same product was obtained via diazotization of **10a** (0.32 g, 0.5 mmol, see below) and stirring overnight with 20 mL of 6% H₂SO₃ at room temperature, yield 0.03 g (10%) after flash column chromatography.

4',4"-Azobis[1-(4-Amino-3,5-diiodobenzyl)-6,7-dimethoxy-1,2,3,4-t trahydroisoquinoline]

A solution of azo-compound **11** (0.26 g, 0.2 mmol) in 35 mL of MeOH and 0.85 g of K2CO3 in 11 mL was refluxed for 4 h and evaporated. Flash chromatography on silica gel (CH₂Cl₂, CH₂Cl₂-MeOH 50:1, 30:1) gave 0.15 g (70%) of the product, mp 176-177°C (dec.); 1H NMR (300 MHz,CDCl₃) δ 7.96 (s, 2H, ArH), 6.62 (s, 1H, ArH), 6.61 (s, 1H, ArH), 4.20 (dd, J = 9.6, 4.0 Hz, 1H, ArH), 3.87 (s, 3H, OMe), 3.86 (s, 3H, OMe), 3.10-3.30 (m, 2H, CH), 3.00 (m, 1H, CH), 2.68-2.90 (m, 2H, CH); ¹³C NMR (CDCl₃) δ 148.43, 147.75, 147.21, 144.05, 141.87, 129.76, 127.43, 112.03, 109.34, 90.43, 56.61, 56.16, 55.88, 41.65, 40.51, 29.36; IR (KBr) 1515 (C=C, Ar) cm⁻¹; Anal. (C₃6H₃6l₄N₄O₄) C, H, N.

EXAMPLE 13

1-(4-Acetamido-3,5-diiodobenzyl)-6,7-dimethoxy-2-trifluoroacetyl-1,2,3,4-tetrahydroisoquinoline

A solution of acetyl chloride (0.80g, 7.8 mmol) in dry THF (2 mL) was added dropwise to a stirred solution of **10**a (1.0g, 1.56 mmol), Et₃N (0.40g, 7.8 mmol), and N, N-dimethylaminopyridine (DMAP, @10 mg) in dry THF (20 mL) at 0 °C under an

argon atmosphere. After the addition the reaction mixture was allowed to warm to room temperature and stirring was continued overnight (14 h). The reaction was quenched with H₂O (20 mL) and stirred for 30 min. The solution was extracted with Et₂OAc (3 x 75 mL). The organic extract was washed with H₂O (20 mL), dried (Na₂SO₄) and evaporated in vacuo to give a tan solid. Recrystallization of the crude product from EtOH and H₂O gave 0.93g (87%) of the title compound as light beige needles: mp 218-219 °C; ¹H NMR (CDCl₃) δ 7.5-7.75 (bm, 2H, ArH), 6.96 (s, 1H, CONH), 6.62 (s, 1H, ArH), 6.24 (s, 1H, ArH), 5.47 (t, J = 6.7 Hz, 1H, CH)3.98 (m, 1H, CH), 3.87 (s, 3H, OMe), 3.73 (s, 3H, OMe), 2.83-3.18 (m, 4H, CH), 2.74, (m, 1H, CH), 2.22 (s, 3H, Ac); ¹³C NMR (CDCl₃) δ 168.14, 156.02, (q), 148.39, 147.68, 140.58, 140.31, 139.35, 125.71, 124.85, 116.4 (q), 98.73, 56.09, 55.93, 55.44, 40.63 (q), 40.48, 28.43, 23.62 IR (KBr) 3387 (NH), 1683 (CO). Anal. (C22H₂1F₃I₂N₂O₄) C, H, N.

EXAMPLE 14

1-(4-Acetamido-3,5-diiodobenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline Hydrochloride

A solution of 13 (1.22g, 1.78 mmol) in methanol (60 mL) was added to a solution of K₂CO₃ (5.6 g) in 80 mL of 1:1 methanol and water. The mixture was stirred at room temperature for 3 hours. The resulting solution was concentrated then extracted with ethyl acetate (3 X 80 mL). The organic solution was dried (Na₂SO₄) and evaporated in vacuo to give the 0.79 g (75%) of the product as the free base. The free base converted to the hydrochloride salt and recrystallized from anhydrous ethanol and ethyl ether: mp 196-200 °C (dec); 1 H NMR (DMSO-D₆) 5 9.85 (s, 1H, CONH), 9.35 (bm, 2H, NH⁺), 7.98 (s, 1H, ArH), 7.96 (s, 1H, ArH), 6.78

(s, 1H, ArH), 6.65 (s, 1H, ArH), 4.65 (bm, 1H, CH), 3.83 (s, 3H, OMe), 3.73 (s, 3H, OMe) 3.35-3.43 (m, 1H, CH), 2.8-3.2 (m, 5H, CH), 2.01 (s, 3H, Me); IR (KBr) 1677 (C=O), 1514 (C=C Ar) cm⁻¹; MS m/e (m+): 592 (M-HCI, EI). Anal. (C₂₀H₂₂I₂N₂O₃+HCI-0.5 Et₂O) C, H, N.

EXAMPLE 15

1-(4-Acetamido-3,5-diiodobenzyl)-6,7-dihydroxy-1,2,3,4-tetrahydroisoquinoline Hydrobromide

To a solution of 14 (0.50 g, 0.88 mmol) in dry CH2Cl2 (50 mL) at 0°C (ice bath) was added dropwise 1M BBr3 (4 mL, 4 mmol) in CH2Cl2 under an argon atmosphere. The mixture was then allowed to reach room temperature and stirring was continued overnight. The reaction mixture was cooled with an ice bath and methanol (20 mL) was added carefully. The solution was stirred for 10 minutes then evaporated in vacuo. This was repeated four times to give a solid which was stirred with ether overnight. The crude product was collected by filtration and recrystallized from methanol and ethyl ether to give 0.51g (90%) of the desired product as an offwhite solid: mp 202-206 (dec) °C; ¹H NMR (DMSO-D₆) δ 9.82 (s, 1H, CONH), 9.15 (bm, 1H, OH), 8.91 (bm, 2H, NH⁺), 8.55 (bm, 1H, OH), 7.97 (s, 1H, ArH), 7.94 (s, 1H, ArH), 6.71 (s, 1H, ArH), 6.56 (s, 1H, ArH), 4.66 (bm, 1H, CH), 3.27-3.35 (m, 2H, CH), 3.10-3.16 (m, 2H, CH), 2.70-2.93 (m, 4H, CH), 2.02 (s, 3H, Me); ¹³C NMR (CD₃OD) δ 172.51(C=O), 147.04 and 145.89 (C-6 and C-7), 142.34 (C-4'), 141.90 and 141.73 (C-2' and C-6'), 140.18 (C-1'), 123.67 and 123.09 (C-4a and C-8a), 116.32 (C-5), 114.11 (C-8), 100.61 and 100.46 (C-3' and C-5'), 57.51 (C-1), 41.10 (C-3), 39.31 (CH₂Ar), 25.70 (C-4) 23.09 (COCH₃); IR (KBr) 1652 (CO), 1524 (C-N) cm-1; MS m/e (m+): 565 (M+ H, FAB). Anal. (C18H18N2O3I2 • HBr • 0.25 Et2O) C, H, N.

1-(4-Diacetamido-3,5-diiodobenzyl)-6,7-dimethoxy-2-trifluoroacetyl-1,2,3,4-tetrahydroisoquinoline

A solution of **10a** (0.70g, 1.08 mmol) in acetic anhydride (10 mL) was heated at reflux for 2h. The solvent was evaporated in vacuo to give an oily residue. The residue was taken up in hot ethanol. The product crystallized upon cooling to give 0.78g (99%) of the product as white crystals: mp 190-192°C; ¹H NMR (CDCl₃) 7.70 (s, 2H, ArH), 6.64 (s, 1H, ArH), 6.39(s, 1H, ArH), 5.53 (t, J = 6.7 Hz, 1H, CH), 3.96 (m, 1H, CH), 3.87 (s, 3H, OMe), 3.79 (s, 3H, OMe), 2.68 (m, 1H, CH), 2.94,-3.07 (m, 3H, CH), 2.77 (m, 1H, CH), 2.28 (s, 6H, Ac); ¹³C NMR (CDCl₃) δ 71.23, 155.93 (q), 148.48, 147.86, 142.66, 141.36, 141.12, 125.65, 124.83, 116.31 (q), 111.14 109.88, 99.21, 56.08, 55.93, 55.31, 40.58, 40.44 (q), 28.43, 26.60; IR (KBr) 1719, 1683 (C=O), 1235, 1207 (C-O) cm⁻¹. Anal. (C24H23F3I2N2O5) C, H, N.

EXAMPLE 17

6,7-Dihydroxy-1-(4-hydroxy-3,5-diiodobenzyl)-1,2,3,4-tetrahydroisoquinoline Hydrobromide

Hydrochloride **17**¹⁰ (0.21 g, 0.28 mmol) was dissolved in CHCl3 and washed with 1N NaOH, organic layer was separated, washed with water and dried over MgSO4. The solution was filtered, evaporated and dried under vacuum. The residue

was dissolved in CH₂Cl₂ (4 mL) and 1M BBr₃ in CH₂Cl₂ (1.39 mL, 1.39 mmol) was added at -78°C under argon atmosphere. The mixture was stirred overnight at room temperature followed by MeOH (1 mL) was added and stirred for 5 h. The resulting solution was evaporated with MeOH 5 times and the residue was recrystallized from MeOH-ether to give 0.078 g (45%) of white crystals : mp 235-237 °C (dec.); ¹H NMR (DMSO-D6) δ 9.50 (s, 1H, OH), 9.14 (s, 1H, OH), 8.89 (s, 1H, OH), 8.78 (br. s, 1H, NH⁺), 8.43 (br. s, 1H, NH⁺), 7.76 (s, 2H, ArH, 6.64 (s, 1H, ArH), 6.55 (s, 1H, ArH), 4.55 (m, 1H, CH), 3.40-3.05 (m, 3H, CH), 2.92-2.68 (m, 3H, CH); ¹³C NMR (CD₃OD) δ 156.59 (C-4'), 146.96 and 145.85 (C-6 and C-7), 141.69 (C-2' and C-6'), 132.30 (C-1'), 123.71 and 123.24 (C-4a and C-8a), 116.25 (C-5), 114.14 (C-8), 85.85 (C-3' and C-5'), 57.55 (C-1), 41.08 (C-3), 39.12 (CH₂Ar), 25.68 (C-4); IR (KBr) 3600-2600 (OH, NH), 1527 (C=C, Ar) cm⁻¹. Anal. (C₁6H₁5l₂NO₃•HBr•0.1Et₂O) C, H, N.

EXAMPLE 18

1-(3,5-Diiodobenzyl)-6,7-dimethoxy-2-trifluoroacetyl-1,2,3,4-tetrahydroisoguinoline

A solution of isoquinoline **10a** (1.29 g, 2 mmol) in glacial acetic acid (30 mL) was added to a cold solution of NaNO₂ (0.19 g, 2.8 mmol) in concentrated (d 1.84) H₂SO₄ (3.4 mL), the temperature was kept within 0-5°C. The solution was poured into ice-water (60g) and H₃PO₂ (12 ml) was added in 30 min. The cooling bath was removed and the solution was allowed to stand at room temperature for 2 days. The precipitate was filtered, dried and chromatographed on silica gel (hexane-AcOEt 8:1). Recrystallization from AcOEt-hexane gave 0.50 g (40%) of white crystals.: mp 162-163 °C; 1H NMR (CDCl₃) δ 7.93 (t, J = 1.5 Hz, 1H, ArH), 7.43 (d, J = 1.5 Hz,

2H, ArH), 6.62 (s, 1H, ArH), 6.24 (s, 1H, ArH), 5.48 (t, J = 6.7 Hz, 1H, CH), 3.95 (m, 1H, CH), 3.87 (s, 3H, OMe), 3.72 (s, 3H, OMe), 3.61 (m, 1H, CH), 3.06- 2.86 (m, 3H, CH), 2.70 (m, 1H, CH); ¹³C NMR (CDCl₃) δ 155.98 (q) 148.46, 147.67 143.60, 141.25, 137.98, 125.86, 125.00, 116.41 (q), 111.20, 110.19, 94.58, 55.96, 55.93, 55.37, 41.13 40.61 (q), 28.44; IR (KBr) 1686 (C=O), 1541, 1520 (C=C Ar) cm-1. Anal. (C18H19l2NO2) C, H, N.

EXAMPLE 19

1-(3,4,5-Triiodobenzyl)-6,7-dimethoxy-2-trifluoroacetyl-1,2,3,4-tetrahydroisoquinoline

Compound **10a** (1.29 g, 2 mmol) was diazotized in the usual manner. The resulting solution was poured in ice-water (60 g) followed by KI (0.47 g, 2.8 mmol) in water (10 mL) was added. The mixture was heated to 80°C and allowed to cool. The precipitate was filtered, dried and purified by column chromatography (silica gel, hexane-AcOEt 3:1). Yield 0.51 g (34 %): mp 215-216 °C; 1 H NMR (CDCl3) 8 7.59 (s, 2H, ArH, 6.63 (s, 1H, ArH), 6.31 (s, 1H, ArH), 5.46 (t, 1 = 6.7 Hz, 1H, ArH), 3.93 (m, 1H, CH), 3.88 (s, 3H, OMe), 3.75 (s, 3H, OMe), 3.61 (ddd, 1 = 13.8, 1 = 9.9, 1 = 4.1 Hz, 1H, CH), 3.02-2.85 (m, 3H, CH₂Ar + CH), 2.70 (dt, 1 = 16.2, 1 = 4.3 Hz, 1H, CH); 13 C NMR (CDCl₃) 8 156.07 (q), 148.48, 147.71, 140.45, 139.90, 125.67, 125.08, 118.91, 116.39 (q), 111.16, 110.08, 106.78, 55.97, 55.10, 40.68 (q), 40.32 28.43; IR (KBr) 1685 (C=O), 1519 (C=C Ar) cm⁻¹. Anal. (C₂0H₁7F₃I₃NO₃) C, H, N.

1-(3,5-Diiod benzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline

A mixture of isoquinoline 19a (0.38 g, 0.6 mmol) in MeOH (35 mL) and K2CO3 (0.85g) in water (11 mL) were refluxed for 1.5 h. MeOH was evaporated and a white precipitate was filtered and dried. The crude product was purified by flash chromatography (silica gel) using a gradient (EtOAc-hexanes 1:2, EtOAc, EtOAc-MeOH 30:1) and recrystallized from EtOAc-hexanes to give 0.25g (76%) of white crystals: mp 122-124 °C; 1 H NMR (CDCl3) δ 7.94 (t, J = 1.4 Hz, 1H, ArH), 7.60 (d, J = 1.4 Hz, 2H, ArH), 6.60 (s, 1H, ArH), 6.57 (s, 1H, ArH), 4.10 (m, 1H, CH), 3.86 (s, 3H, OMe), 3.84 (s, 3H, OMe), 3.18 (m, 1H, CH), 3.08 (dd, J = 4.1 and 13.7 Hz, 1H, CH), 2.95 (m, 1H, CH), 2.82-2.61 (m, 3H, CH); 13 C NMR (CDCl3) δ 147.68, 147.17, 143.78, 143.15, 137.64, 129.86, 127.42, 111.98, 109.32, 94.99, 56.57, 56.06, 55.87, 42.20, 40.55, 29.39; IR (KBr) 3325 (NH), 1516 (C=C Ar) cm⁻¹ Anal. (C18H19l2NO2) C, H, N.

EXAMPLE 21

6,7-Dimethoxy-1-(3,4,5-triiodobenzyl)-1,2,3,4-tetrahydroisoquinoline

A mixture of isoquinoline 19b (0.454 g, 0.6 mmol) in MeOH (35 mL) and K2CO3 in water (11 mL) were refluxed for 1.5 h. MeOH was evaporated and a white precipitate was filtered and dried. Recrystallization from CHCl3-hexane gave 0.300 g (76 %) of white crystals: mp 168-170 °C (dec.); 1 H NMR (CDCl3) 8 7.79 (s, 2H, ArH), 6.60 (s, 1H, ArH), 6.58 (s, 1H, ArH), 4.09 (dd, 1 J = 9.8, 3.8 Hz, 1H, CH), 3.86 (s, 3H, OMe), 3.85 (s, 3H, OMe), 3.17 (m, 1H, CH), 3.04-2.88 (m, 2H, CH), 2.82-2.60 (m, 3H, CH); 13 C NMR (CDCl3) 8 147.66, 147.15, 143.00, 139.68, 129.67, 127.42, 118.18, 111.92, 109.14, 107.09, 56.35, 56.06, 55.85, 41.38, 40.56, 29.35; IR (KBr) 3312 (NH), 1516 (C=C Ar) cm⁻¹. Anal. (C18H18I3NO2) C. H. N.

EXAMPLE 22

1-(4-Amino-3,5-diiodobenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoguinoline

A slurry of the isoquinoline **10a** (1.94 g, 3 mmol) in MeOH (260 mL) and K2CO3 (11.2 g) in H2O (80 mL) was refluxed for 1 h. MeOH was evaporated under reduced pressure, crystals filtered, dried and recrystallized from EtOAc-hexane to give the title compound (1.39 g, 84 %): mp 169-171 °C; ¹H NMR (CDCl₃) δ 7.55 (s, 2H, ArH), 6.61 (s, 1H, ArH, 6.51 (s, 1H, ArH), 4.54 (s, 2H, NH₂), 4.04 (dd, J = 9.6, 4.0 Hz, 1H, CH), 3.86 (s, 3H, OMe), 3.84 (s, 3H, OMe), 3.18 (m, 1H, CH), 3.03 (dd, J = 13.8, 4.0 Hz, 1H, CH), 2.92 (ddd, J = 12.1, 6.8, 5.2 Hz, 1H, CH), 2.82-2.61 (m, J = 13.8, 9.6 Hz, 3 H, CH); ¹³C NMR (CDCl₃) δ 147.51, 147.05, 144.66, 139.97, 132.36, 130.14, 127.38, 111.88, 109.28, 81.59, 56.79, 56.01, 55.83, 40.87, 40.72, 29.47; IR (KBr) 3416 (NH), 3331 (NH), 1607 (NH bend), 1571, 1512 (C=C Ar) cm⁻¹. Anal. (C18H20N2I2) C, H, N.

6,7-Dihydroxy-1-(3,5-diiodobenzyl)-

1,2,3,4-tetrahydroisoquinoline Hydrobromide

The isoquinoline **20a** (0.19 g, 0.35 mmol) was demethylated using the same procedure as **15**. Recrystallization from MeOH-ether gave 0.20 g (96%) of the title compound: mp 157-159 °C (dec.); 1 H NMR (DMSO-D6) 8 9.13 (bs, 1H, OH), 8.88 (bm, 2H, NH+OH), 8.57 (bm, 1H, NH), 8.02 (t, J = 1.4 Hz, 1H, ArH), 7.80 (d, J = 1.4 Hz, 2H, ArH), 6.61 (s, 1H, ArH), 6.56 (s, 1H, ArH), 4.63 (bm, 1H, CH), 3.22-3.41 (m, 2H, CH), 3.14 (m, 1H, CH), 2.71-2.96 (m, 3H, CH), 13 C NMR (CD3OD) 8 147.01 and 145.79 (C-6 and C-7), 145.69 (C-4'), 139.18 (C-2' and C-6'), 141.15 (C-1'), 123.76 and 123.03 (C-4a and C-8a), 116.30 (C-5), 114.21 (C-8), 96.14 (C-3' and C-5'), 57.25 (C-1), 41.02 (C-3), 40.08 (CH2Ar), 25.60 (C-4); IR (KBr) 3600-2700 (br. OH, NH). 1617, 1521 (C=C Ar) cm⁻¹. Anal. (C16H15Brl3NO2) C, H, N.

EXAMPLE 24

6,7-Dihydroxy-1-(3,4,5-triiodobenzyl)1,2,3,4-tetrahydroisoquinoline Hydrobromide

The isoquinoline **20b** (0.23 g, 0.35 mmol) was demethylated using the same procedure as **15**. Recrystallization from MeOH-ether gave 0.24 g (97%) of the title compound: mp 210-213 °C (dec.); ¹H NMR (MeOH-D4) δ 7.92 (s, 2H, ArH), 6.63 (s,

1H, ArH), 6.56 (s, 1H, ArH), 4.64 (dd, J = 5.7 and 3.1 Hz, 1H, CH), 3.3.42-3.53 (m, 1H, CH), 3.2-3.34 (m, 2H, CH), 2.83-3.07 (m, 3H, CH), 13 C NMR (CD3OD) δ 147.11 and 145.90 (C-6 and C-7), 141.06 (C-2' and C-6'), 140.21 (C-1'), 123.70 and 122.98 (C-4a and C-8a), 120.90 (C-4'), 116.31 (C-5), 114.14 (C-8), 108.68 (C-3' and C-5'), 57.04 (C-1), 41.01 (C-3), 39.38 (CH₂Ar), 25.62 (C-4); IR (KBr) 3600-2700 (br. OH, NH). 1617, 1540 (C=C Ar) cm⁻¹. Anal. (C₁₆H₁₆Brl₂NO₂) C, H, N.

EXAMPLE 25

1-(4-Amino-3,5-diiodobenzyl)-6,7-dihydroxy-1,2,3,4-tetrahydroisoquinoline Dihydrochloride

The isoquinoline **20c** (1.21 g, 2.2 mmol) was demethylated in the same manner as **20b** to give 1.14 g (76 %) of the dihydrobromide salt: mp 155-157°C (dec.). The product was dissolved in MeOH, chromatographed (silica gel, EtOAc - NH4OH 100:1), evaporated with EtOH (x 5). To an ethanol solution was added 1N etherial solution of HCl (3 mL), concentrated, precipitated with EtOAc and recrystallized from MeOH-i-PrOH: mp 176-178 °C (dec.); ¹H NMR (DMSO-D6) δ 9.15 (br s, 1H, OH), 8.89 (br.s, 1H, NH2+), 7.68 (s, 2H, H-2'), 6.64 (s, 1H, H-5), 6.55 (s, 1H, H-8), 5.06 (br s, 2H, NH2), 4.47 (m, 1H, H-1), 3.40-2.67 (m, 6H, H-3 + H-4 + CH2Ar); ¹³C NMR (CD3OD) δ 147.99 (C-4'), 146.84 and 145.75 (C-6 and C-7), 141.60 (C-2' and C-6'), 128.78 (C-1'), 123.75 and 123.33 (C-4a and C-8a), 116.24 (C-5), 114.16 (C-8), 81.86 (C-3' and C-5'), 57.59 (C-1), 41.07 (C-3), 39.07 (CH2Ar), 25.68 (C-4); IR (KBr) 3600-2500 (br, OH, NH), 1607 (NH bend), 1529 (C=C Ar) cm⁻¹. Anal. (C16H16l2N2O2*2HCl*H2O) C, H, N.

1-(4-Acetamido-3-iodobenzyl)-6,7-dimethoxy-2-trifluoroacetyl-1,2,3,4-tetrahydroisoquinoline

To a solution of isoquinoline 9a (0.52 g, 1 mmol) in hot benzene (15 mL) was added Ac₂O (0.51 g, 5 mmol). The solution was refluxed for 1 h. Reaction mixture was allowed to cool. A white crystals were filtered. Mother liquor was concentrated and hexane was added. Slightly creamy crystals were filtered, total yield 0.53 g (94 %). To get analytical sample the compound was recrystallized from EtOAc - hexane: mp 174-175 °C; 1 H NMR (CDCl₃) 8 8.11 (d, 1 9 = 8.3 Hz, 1H, H-5'), 7.52 (d, 1 9 = 1.5 Hz, 1H, H-2'), 7.36 (s,1H, NH), 7.10 (dd, 1 9 = 8.3, 1.5 Hz, 1H, ArH), 6.60 (s, 1H, ArH), 6.32 (s, 1H, ArH), 5.23 (t, 1 9 = 6.6 Hz, CH), 3.94 (m, 1H, CH), 3.87 (s, 3-H, OMe), 3,72 (s, 3-H, OMe), 3.54 (ddd, OMe = 14.1, 10.4 Hz, 3.8 Hz,1H, CH), 3.06 (m, 2H, CH), 2.90 (ddd, 1 9 = 15.9 Hz, 10.4 Hz, 5.2 Hz, 1H, CH), 2.68 (dt, 1 9 = 16.0, 4 Hz, 1H, CH), 2.23 (s, 3H, Ac); 13 C NMR (CDCl₃) 8 9 168.12, 155.90 (q), 148.30, 147.63, 139.63, 137.12, 134.95, 130.53, 126.19, 124.95, 121.73, 116.44 (q), 111.10, 110.17, 89.68, 55.91, 55.33, 40.75, 40.50 (q), 28.51, 24.75; IR (KBr) 3395 (NH), 1688 (C=O), 1519 (C=C Ar) cm⁻¹. Anal. (C22H22F3IN2O4) C, H, N.

EXAMPLE 27

1-(4-Acetamido-3-iodobenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline

The title compound (0.57 g, 69%) as a glassy solid was obtained from the isoquinoline **22** (0.99 g, 1.76 mmol) in the same manner as **20c**. ¹H NMR (CDCl₃) δ 8.12 (d, J = 8.3 Hz, 1H, H-5'), 7.70 (d, J = 1.7 Hz, 1H, ArH), 7.38 (s, 1H, NH), 7.26 (dd, J = 8.3 Hz, 1.7 Hz, 1H, ArH), 6.63 (s, 1H, ArH), 6.60 (s, 1H, ArH), 4.11 (dd, J = 9.6 Hz, 3.8, 1H, CH), 3.10-3.25 (m, 2H, CH), 2.92 (m, 1H, CH), 2.63-2.86 (m, 3H, CH₂Ar, CH), 2.25 (s, 3H, Ac); ¹³C NMR (CDCl₃) δ 168.19, 147.54, 147.05, 139.31, 137.30, 136.66, 130.12, 129.92, 127.28, 122.25, 111.85, 109.27, 90.50, 56.64, 55.99, 55.79, 41.63, 40.61, 29.30, 24.66; IR (KBr) 3391 (NH), 1676 (C=O), 1515 (C=C Ar) cm⁻¹. Anal. (C₂0H₂3IN₂O₃) C, H, N.

EXAMPLE 28

1-(4-Amino-3-iodobenzyl)-6,7-dihydroxy-1,2,3,4-tetrahydroisoquinoline Dihydrochloride

The title compound was obtained in the same manner as **21c** from the isoquinoline **23** (0.42 g, 0.90 mmol). The product was recrystallized from MeOH (twice) to give 0.32 g (64%). The compound was dissolved in NaHCO3 solution and extracted with EtOAc (x 5). The solution was dried (MgSO4) and evaporated. 1M HCI in ether (2 mL) was added to the methanol solution of the residue. The solution was concentrated and put in a refrigerator. The white crystals were filtered, washed with EtOAc and dried: dec.p. 186-190 °C; ¹H NMR (DMSO-D6) δ 9.05 (bs, 1H , OH), 7.68 (d, J = 1.7 Hz, 1H, ArH), 7.16 (dd, J = 8.2 Hz, 1.7 Hz, 1H, ArH), 6.94 (d, J = 8.2 Hz, 1H, ArH), 6.57 (s, 1H, ArH), 6.55 (s, 1H, ArH), 4.46 (m, 1H, CH), 3.27 (m, 1H, CH), 3.00-3.20 (m, 2H, CH), 2.80-3.00 (m, 2H, CH), 2.74 (dt, J = 16.8 Hz, 5.9 Hz, CH); 13C NMR (CD3OD) δ 147.00 and 145.76 (C-6 and C-7), 1412.68 (C-2'), 138.21 (C-4'), 136.27 (C-1'), 132.35 (C-6'), 123.82 and 123.10 (C-4a and C-8a), 116.27 (C-5), 114.33 (C-8), 124.24 (C-5'), 91.49 (C-3'), 57.27 (C-1), 40.90 (C-3), 39.87

(CH₂Ar), 25.659 (C-4); IR (KBr) 3600-2300 (br, OH, NH), 1607 (NH bend), 1526 (C=C Ar) cm⁻¹. Anal. (C₁₆H₁₇IN₂O₂•2HCl) C, H, N.

EXAMPLE 29

1-(4-Acetamido-3-iodobenzyl)-6,7-dibenzyloxy-2-tert-butoxycarbonyl-1,2,3,4-tetrahydroisoquinoline

To a cold solution (0° C) of isoquinoline **9b** (0.68 g, 1 mmol) and Et₃N (0.34 g, 3 mmol) in CH₂Cl₂ (10 mL) was added AcCl (0.16 g, 2 mmol). The cooling bath was removed and the mixture was stirred overnight. The solution was washed with water (2x), dried over MgSO₄, filtered, evaporated. Ether was added and evaporated again to give glassy solid (0.65 g, 90%): mp 62-64 °C; ¹H NMR (CDCl₃) δ (the spectrum consists of 2 rotamers of 5:3 ratio) 8.12 and 8.06 (d, J = 8.2), 7.59-7.25 (m, 11H, ArH), 7.06 and 6.98 (m, 1H, ArH), 6.70 and 6.65 (s, 1H, ArH), 6.48 and 6.35 (s, 1H, ArH), 5.24-4.87 (m, 5H, CH₂O + CH), 4.12 and 3.73 (m, 1H, CH), 3.27-3.11 (m, 1H, CH), 2.98-2.60 (m, 3H, CH₂Ar + CH), 2.60-2.37 (m, 1H, CH), 2.22 (s, 3H, Ac), 1.43 and 1.31 (s, 9H, t-Bu); IR (KBr) 3389 (NH), 1688 (C=O), 1512 (C=C Ar) cm⁻¹. Anal. (C₃7H₃9IN₂O₅) C, H, N.

EXAMPLE 30

1-(4-Benzoylamino-3-iodobenzyl)-6,7-dibenzyloxy-2-tert-butoxycarbonyl-1,2,3,4-tetrahydroisoquinoline

To a cold solution (0°C) of isoquinoline 9b(0.68 g, 1 mmol) and Et₃N (0.30 g, 3 mmol) in 10 mL of CH₂Cl₂ was added benzoyl chloride (0.28 g, 2 mmol). the cooling bath was removed and the mixture was stirred overnight. CH₂Cl₂ was added (30 mL), the solution was washed with water, dried over MgSO₄, filtered and evaporated till dryness. The oily residue was dissolved in ether and evaporated to give 0.60 g (76 %) of a glassy solid. The compound was purified by column chromatography (silica gel, EtOAc-hexane 1:2): mp 151-153 °C; ¹H NMR (CDCl₃) δ 8.42-6.36 (m, 18 H, Ar), 5.20-4.90 (m, 5H, 2 x CH₂O + H-1), 4.20-2.15 (m, 6H, aliphatic), 1.56-1.25 (m, 9H, t-Bu); IR (KBr) 3397 (NH), 1687 (C=O), 1513 (C=C Ar) cm⁻¹. Anal. (C42H41IN₂O₅) C, H, N.

EXAMPLE 31

1-(4-Acetamido-3-iodobenzyl)-6,7-dihydroxy-1,2,3,4-tetrahydroisoquinoline Hydroiodide

To a solution of isoquinoline 25a (0.40 g, 0.5 mmol) in anhydrous MeCN (5 mL) was added TMSI (0.40 g, 2 mmol) via syringe in argon atmosphere. The solution was stirred for 6h followed by MeOH (1 mL) was added and stirring continued for 30 min. CH2Cl2 (30 mL) was added to reaction mixture and yellow crystals were filtered, yield 0.19 g (67 %). The compound was dissolved in MeOH. AcOEt was added and the solution was concentrated under reduced pressure. The crystals were filtered: dec.p. 172-174 °C; ¹H NMR (DMSO-D6) δ 9.39 (s, 1H, NH), 8.86 (bs, 1H, OH), 8.50 (bs, 1H, OH), 7.90 (d, J = 1.7 Hz, 1H, ArH), 7.41 (d, J = 8.2Hz, 1H, ArH), 7.35 (dd, J = 8.2, 1.7 Hz, 1H, ArH), 6.63 (s, 1H, ArH), 6.56 (s, 1H, ArH), 4.63 (m, 1H, CH), 3.43-2.70 (m, 6H, CH), 2.06 (s, 3H, Ac); ¹³C NMR (CD₃OD) δ 172.65 (C=O), 146.98 and 145.81(C-6 and C-7), 142.68 (C-2'), 140.11 (C-4'), 137.15 (C-1'), 131.30 (C-6'), 129.10 (C-5'), 123.63 and 123.27 (C-4a and C-8a), 116.27 (C-5), 114.15 (C-8), 91.49 (C-3'), 57.60 (C-1), 41.01 (C-3), 39.98 (CH₂Ar). 25.68 (C-4), 23.09 (COCH3); IR (KBr) 3600-2400 (br, OH, NH), 1655 (C=O), 1624 (NH bend), 1522 (C=C Ar) cm⁻¹; MS m/z (m⁺); 439. Anal. (C₁₈H₁₉IN₂O₃•HI•0.25 EtOAc) C, H, N.

EXAMPLE 32

1-(4-Benzoylamino-3-iodobenzyl)-6,7-dihydroxy-1,2,3,4-tetrahydroisoquinoline Hydroiodide

To a mixture of isoquinoline **25b** (0.16 g, 2 mmol) in MeCN (4 mL) was added TMSI (0.16 g, 0.8 mmol) under argon atmosphere and the solution was stirred at room temperature for 7 h. MeOH (1 mL) was added, stirred for 1 h followed by ether (40 mL) was added and the yellow precipitate was filtered to give 0.10 g (80 %) of the product. The compound was dissolved in MeOH, EtOAc was added and

concentrated until the beginning of crystallization: mp 185-188 °C (dec.); ¹H NMR (DMSO-D6) δ 9.98 (s, 1H, NHCOPh), 9.18 (s, 1H, NH), 8.88 (br, 2H, OH + NH), 8.54 (br, 1H, OH), 8.03-7.93 (m, 3H, ArH), 7.64-7.42 (m, 3H, ArH), 7.49 (d, J = 8.1 Hz, 1H, ArH), 7.41 (dd, J = 8.1, 1.4 Hz, 1H, ArH), 6.65 (s, 1H, ArH), 6.58 (s, 1H, ArH), 3.43-2.70 (m, 6H, CH; IR (KBr) 3500-2700 (br, NH, OH), 1649 (C=O), 1518 (C=C Ar) cm⁻¹. Anal. (C₂₃H₂1IN₂O₃+HI+0.33 EtOAc) C, H, N.

EXAMPLE 33

1-(3,5-Bis-trifluoromethylbenzyl)-6,7-dihydroxy-1,2,3,4-tetrahydroisoquinoline Hydrochloride

The title compound was obtained from **6c** in the same manner as **15**. The product was converted to the hydrochloride salt and recrystallization from methanolether gave the product as white crystals: mp 239-242 °C; ¹H NMR (CD₃OD-D₄) δ 7.95 (s, 3H, ArH), 6.65 (s, 1H, ArH), 6.44 (s, 1H, ArH), 4.77 (t, J = 7.7 Hz), 3.51 (dt, J = 6.88, 12.74 Hz), 3.33 (m, 2H, CH), 3.0 (m, 2H, CH), ¹³C NMR (CD₃OD) δ 147.15 and 145.81 (C-6 and C-7), 140.21 (C-1'), 133.21 (C-3' and C-5'), 131.61 (C-2' and C-6'), 124.79 (CF₃), 123.79 and 122.63 (C-4a and C-8a), 122.58 (C-4'), 116.40 (C-5), 114.35 (C-8), 57.11 (C-1), 41.77 (C-3), 40.55 (CH₂Ar), 25.56 (C-4); IR (KBr) 3420 (NH), 1282 (C-O) cm⁻¹; Anal. (C₁₈H₁₆ClF₆NO₂) C, H, N.

EXAMPLE 34

The compounds of the foregoing examples 1 to 33 were subjected to elemental analysis to further support the structural determinations. The results are summarized in Table I:

Table I Elemental Analyses

	Calcu	ılated,	%		Foun	d, %	
Example	С	Н	N	Formula	C	Н	N
5a	62.7	5.85	8.13	C ₁₈ H ₂₀ N ₂ O ₅	62.6	5.84	8.05
E.o.	8	4.40	0.00	CoalliaFaNO	2		
5c	55.1 7	4.40	3.22	C ₂₀ H ₁₃ F ₆ NO ₃	. 55.1 0	4.41	3.20
6a	65.8 4	6.14	8.53	C ₁₈ H ₂₀ N ₂ O ₄	65.1	6.17	8.46
6c	51.6 8	4.55	3.01	C ₂₀ H ₁₉ F ₆ NO ₂ • HCl • 0.5H ₂ O	9 51.7	4.53	2.99
7a	56.6 0	4.51	6.60	C ₂₀ H ₁₉ F ₃ N ₂ O ₅	0 56.5	4.52	6.66
7b	64.4 7	6.59	6.54	C ₂₃ H ₂₈ N ₂ O ₆	9 64.5 7	6.66	6.48
7c	72.4 0	6.25	4.82	C35H36N2O6	72.3 2	6.26	4.79
8a	60.9 1	5.37	7.10	C ₂₀ H ₂₁ F ₃ N ₂ O ₃	60.9	5.39	7.18
8b	69.3 2	7.59	7.03	C ₂₃ H ₃₀ N ₂ O ₄	69.2 1	7.61	7.05
8c	76.3 4	6.96	5.09	C35H38N2O4	76.2 2	6.98	5.03
9a	46.1 7	3.87	5.38	C ₂₀ H ₂₀ F ₃ IN ₂ O ₃	46.4 0	3.90	5.32
9b	52.6 8	5.57	5.34	C ₂₃ H ₂ 9IN ₂ O ₄	52.6 1	5.56	5.25
9c	62.1 3	5.51	4.14	C35H37IN2O4	62.0 5	5.50	4.04
10a	37.1 8	2.96	4.34	C ₂₀ H ₁₉ F ₃ N ₂ O ₃	37.4 5	3.05	4.32
11	37.2 9	2.66	4.35	C40H34F6I4N4O6		2.77	4.34
12	39.4 4	3.31	5.11	C36H34I4N4O4	39.5 8	3.30	5.09
13	38.4 0	3.08	4.07	C ₂₂ H ₂₁ F ₃ I ₂ N ₂ O ₄	38.5 0	3.27	4.05
14	39.6 9	4.24	4.21	C ₂₀ H ₂₂ I ₂ N ₂ O ₃ • HCI • 0.5 Et ₂ O	39.6 4	3.94	4.44

15	34.3 6	3.25	4.23	C ₁₈ H ₁₈ I ₂ N ₂ O ₃ • HBr • 0.25 Et ₂ O	34.2 7	3.34	4.23
16	39.4 7	3.17	3.84	C ₂₄ H ₂₃ F ₃ l ₂ N ₂ O ₅	39.4 1	3.16	3.85
18	32.2 2	2.80	2.29	C ₁₆ H ₁₅ I ₂ NO ₃ • HBr • 0.1 Et ₂ O	32.3 0	2.76	2.33
19a	38.0 6	2.87	2.22	C ₂₀ H ₁₈ F ₃ I ₂ NO ₃	38.1 1	2.93	2.21
19b	31.7 3	2.26	1.85	C ₂₀ H ₁₇ F ₃ I ₃ NO ₃	31.8 7	2.32	1.84
20a	40.4 0	3.58	2.62	C ₁₈ H ₁₉ I ₂ NO ₂	40.5 2	3.65	2.59
20b	32.7 0	2.74	2.12	C ₁₈ H ₁₈ I ₃ NO ₂	32.5 8	2.73	2.05
20c	39.3 0	3.66	5.09	C ₁₈ H ₂₀ l ₂ N ₂ O ₂	39.5 6	3.73	5.00
21a	32.6 8	2.74	2.38	C ₁₆ H ₁₅ I ₂ NO ₂ • HBr	32.7 7	2.78	2.33
21b	26.9 2	2.12	1.96	C ₁₆ H ₁₄ I ₃ NO ₂ • HBr	27.0 6	2.19	1.91
21c	31.3 5	3.29	4.57	C ₁₆ H ₁₆ I ₂ N ₂ O ₂ • 2 HCl • H ₂ O	31.4 9	3.19	4.35
22	48.9 9	4.11	4.76	C22H22F3IN2O4 • 0.33 PhH	49.0 5	4.10	4.72
23	51.5 1	4.97	6.01	C ₂₀ H ₂₃ IN ₂ O ₃	51.7 3	5.00	5.98
24	40.9 6	4.08	5.97	C ₁₆ H ₁₇ IN ₂ O ₂ • 2 HCl	41.0 4	4.13	5.94
25a	61.8 4	5.47	3.90	C37H39IN2O5	61.9 1	5.46	3.94
25b	64.6 2	5.29	3.59	C42H41IN2O5	64.7 1	5.32	3.67
26a	38.8 0	3.77	4.76	C ₁₈ H ₁₉ IN ₂ O ₃ • 0.25 EtOAc	38.6 5	3.87	4.57
26b	44.4 4	3.78	4.26	C ₂₃ H ₂₂ IN ₂ O ₃ • 0.33 EtOAc	44.6 8	3.79	4.40

Thiazolopyridine Derivatives

2-Amino-4-(2-phthalimidoethyl)thiazole Hydrobromide(13). To a solution of **12** (2.08 g, 7.0 mmol) in acetone (45 ml) was added a solution of thiourea (0.535 g, 7.0 mmol) in acetone (25 ml) with rapid rate at room temperature. Just after the addition was complete, a precipitate appeared, the suspension was stirred overnight at room temperature and filtered to afford 2.45 g (98.5%) of **13** as a colorless powder: m.p. 258-260 °C (dec) (lit. 195 °C (dec starting point)); ¹H NMR (DMSO-d₆), 2.82 (t, J = 6.3 Hz, 2H), 3.83 (t, J = 6.2 Hz, 2H), 6.56 (s, 1H), 7.81-7.88 (m, 4H), 9.01 (bs, 2H); IR (KBr) 3214, 3092, 1767, 1721, 1633, 1573, 1402 crn⁻¹. Anal. ($C_{13}H_{12}N_3O_2SBr$): C, H, N.

N-2-(2'-amino-4'-thiazolyl)ethyl-3,4,5-trimethoxyphenylacetamide (17). (a) To a suspension of 3,4,5-trimethoxyphenylacetic acid (4.52 g, 0.02 mol) in dry benzene (200 ml) was added dropwise oxalyl chloride (20 ml, 0.23 mol) at 0 °C. After the addition was complete, the reaction mixture was stirred at room temperature until a clear solution was obtained (about 1 h). The solution was then heated at reflux for 2.5 h. The reaction mixture was cooled to room temperature and evaporated to give a yellow oil. It was dissolved in benzene (~50 ml) and evaporated again (repeated for two more times). 4.9 g (100%) of the acid chloride was obtained as a viscous yellow oil after dried in vacuo. ¹H NMR (CDCl₃), δ 3.83 (s, 3H), 3.84 (s, 6H), 4.05 (s, 2H), 6.45 (s, 2H); IR (neat) 300, 2941, 2840, 1799, 1593 cm⁻¹. (b) To a well-stirred suspension of 14 (976 mg, 3.2 mmol), NaOH (512 mg, 12.8 mmol) in CHCl₃ (7 ml) and H₂O (5 ml) was added slowly a solution of above-obtained 3,4,5-trimethoxyphenylacetyl chloride (784 mg, 3.2 mmol) in CHCl₃ (6 ml) at room temperature. After the addition was complete, the reaction mixture was stirred vigorously at room temperature for 1.5 h. The CHCl₃ layer was separated and the H₂O layer was extracted with CHCl₃. The combined organics were dried over anhydrous Na₂SO₄, and concentrated under reduced pressure. The oily residue was dissolved in CHCl₃ (40 ml), and to which HCl (1.0 M solution in dry Et₂O) (10 ml) was added at 0 °C. The whole mixture was concentrated, the oily residue was dissolved in H₂O (10 ml). The aqueous solution was washed successively with EtOAc, Et₂O, CHCl₃ and basified with 20% NaOH. The product

was extracted with CHCl₃, the combined organics were washed with brine, dried over anhydrous Na₂SO₄ and concentrated, the resulting solid residue was crystallized from EtOAc to afford 803 mg (71.4%) of **17** as a colorless crystal: m.p. 148-149.5 °C; ¹H NMR (CDCl₃, δ 2.57 (t, J = 6.2 Hz, 2H), 3.42 (q, J = 5.8 Hz, 2H), 3.46 (s, 2H), 3.79 (s, 3H), 3.80 (s, 6H), 5.16 (bs, 2H), 5.96 (s, 1H), 6.43 (s, 2H), 6.50 (bs, 1H); ¹³C NMR (CDCl₃), δ 30.64, 38.88, 44.08, 56.11, 60.78, 103.10, 106.78, 130.53, 136.90, 150.07, 153.29, 167.89, 170.71; IR (KBr) 3433, 3266, 3080, 2939, 1646, 1624, 1590 cm⁻¹. Anal. (C₁₆H₂₁N₃SO₄): C, H, N.

2-Amino-4-(3',4',5'-trimethoxyphenylmethyl)-4,5,6,7-tetrahydrothiazolo[5,4-c]pyridine Dihydrochloride (7). A mixture of 17 (70.3 mg, 0.2 mmol) and phosphorus oxychloride (0.27 ml, 2.9 mmol) in CH₃CN (4 ml) was stirred and heated at reflux for 5 h. After the reaction mixture was cooled and concentrated under reduced pressure, the residue was dissolved in MeOH (2 ml) and the solution was heated at reflux for 30 min.. After evaporation, the residue was dissolved in MeOH and evaporated again (repeated three more times). To the stirred solution of the resulting residue in MeOH (10 ml) was added NaBH₄ (757 mg, 20 mmol) in portions cautiously at 0 °C. After the addition was complete, the reaction mixture was stirred overnight at room temperature. After the reaction mixture was evaporated to dryness under reduced pressure, the residue was dissolved in H₂O (5 ml), cooled with an ice-water bath and was basified with 20% NaOH. The basic solution was extracted with EtOAc, the combined organics were washed with brine, dried over anhydrous Na₂SO₄ and concentrated to give a viscous yellow oil. The oily residue was dissolved in CHCl₃ (5 ml), and to which HCl (1.0 M solution in dry Et₂O) (2 ml) was added at 0 °C. The precipitate was filtered off and washed successively with Et₂O, EtOAc, CHCl₃ and crystallized from MeOH-Et₂O to afford 36.5 mg (44.7%) of 7 as a pale yellow powder: m.p. 230-231 °C (dec); 1H NMR (DMSO-d₆), δ 2.71-2.90 (m, 2H), 2.96-3.03 (m, 1H), 3.17-3.23 (m, 3H), 3.64 (s, 3H), 3.75 (s, 6H), 4.77(bs, 1H), 6.67 (s, 2H), 8.48 (bs, 1H), 9.68 (bs, 1H), 9.85 (bs, 1H); 13 C NMR (CD₃OD), δ 21.78, 39.79, 41.02, 54.83, 56.84, 61.17, 108.31, 112.94, 130.73, 133.85, 139.15, 155.19, 171.53; IR (KBr) 3392, 2940, 2839, 2771, 1632, 1593 cm⁻¹. Anal. $(C_{16}H_{23}N_3SO_3CI_2 \cdot 0.5H_2O)$: C, H, N.

N-2-(2'-Amino-4'-thiazolyl)ethyl-3,5-diiodo-4-methoxyphenylacet- amide (18). In the same manner as 17, the title compound was prepared from 14 (590.3 mg, 1.94 mmol) and 3,5-diiodo-4-methoxyphenylacetyl chloride (844.8 mg, 1.94 mmol) which in turn was obtained by treating its corresponding acid[36] with oxalyl chloride as described above for 3,4,5-trimethoxyphenylacetic acid. Recrystallization from EtOAc gave 642.8 mg (61.0%) of 18 as a colorless crystal: m.p. 175-176 °C; ¹H NMR (CD₃OD), δ 2.63 (t, J = 6.8 Hz, 2H), 3.35 (s, 2H), 3.41 (t, J = 6.8 Hz, 2H), 3.79 (s, 3H), 6.05 (t, J = 0.8 Hz, 1H), 7.72 (s, 2H); ¹³C NMR (CD₃OD), δ 31.87, 39.72, 41.80, 61.14, 90.93, 103.35, 137.10, 141.64, 149.90, 159.38, 171.47, 172.89; IR (KBr) 3431, 3272, 3083, 2933, 1643, 1623, 1579, 1524 cm⁻¹. Anal. (C₁₄H₁₅N₃SO₂I₂): C, H, N.

2-Amino-4-(3',5'-diiodo-4'-methoxyphenylmethyl)-4,5,6,7-tetrahydr-othiazolo[5,4-c]pyridine Dihydrochloride (9). In the same manner as **7**, the title compound was prepared from **18** (380.1 mg, 0.7 mmol). After flash column chromatography on silica gel, eluting with MeOH/CHCl₃ (1:30), 72 mg (19.5%) of the free base form of the product was obtained as a white solid, it was treated with HCl (1.0 M solution in dry Et₂O) to afford **9** as a pale yellow solid: m.p. 201-203 °C (dec); ¹H NMR (DMSO-d₆), δ 2.67-2.77 (m, 2H), 3.02-3.11 (m, 2H), 3.15-3.35 (m, 2H), 3.74 (s, 3H), 4.74 (bs, 1H), 7.90 (s, 2H), 8.31 (bs, 2H), 9.59 (bs, 1H), 9.79 (bs, 1H); ¹³C NMR, δ 22.45, 38.02, 41.48, 54.86, 61.21, 91.85, 112.55, 135.12, 135.46, 142.57, 160.65, 171.36; IR (KBr) 3421, 2964, 2937, 2775, 1628, 1577 cm⁻¹. Anal. (C₁₄H₁₇N₃SOCl₂l₂): C, H, N.

N-2-(2'-Acetamido-4'-thiazolyl)ethyl-3,5-diiodo-4-methoxyphenyl-acetamide (19). To a stirred suspension of 18 (434.5 mg, 0.8 mmol) in dry CH₃CN (1.8 ml) was added dropwise a solution of acetic anhydride (o.16 ml, 1.7 mmol) in dry benzene (0.6 ml). After the addition was complete, the reaction mixture was heated at reflux for 2.5 h. After the reaction mixture was cooled to room temperature and concentrated under reduced pressure to remove solvents completely, H₂O (5 ml) was added to the residue and the mixture was basified with saturated NaHCO₃ aqueous solution to pH 7.5-8.0. The solid material was filtered off and crystallized from CH₃CN to afford 430 mg (91.8%) of 19 as a colorless crystal: m.p. 231-232 °C; ¹H NMR(DMSO-d₆), δ 2.10 (s, 3H), 2.69 (t, J = 7.0 Hz, 2H), 3.29-3.32 (m, 4 H), 3.71

(s, 3H), 6.70 (s, 1H), 7.67 (s, 2H), 8.08 (t, J = 5.4 Hz, 1H), 12.04 (s, 1H); 13 C NMR (DMSO-d₆), δ 22.45, 31.09, 38.17, 40.12, 60.23, 90.94, 107.97, 136.65, 139.93, 148.29, 156.96, 157.52, 168.19, 169.32; IR (KBr) 3429, 3273, 3062, 1644, 1554, 1537 cm⁻¹. Anal. (C₁₆H₁₇N₃SO₃I₂): C, H, N.

2-Acetamido-4-(3',5'-diiodo-4'-methoxyphenylmethyl)-4,5,6,7-tetrahydrothiazolo[5,4-c]pyridine Maleate (10). In the same manner as 7, the title compound was prepared from 19 (300 mg, 0.51 mmol). After flash column chromatography on silica gel, eluting with acetone/hexane (1:3), 150 mg (51.7%) of the free base form of 10 was obtained as a white powder: m.p. 170-172 °C; ¹H NMR (DMSO- d_6), δ 2.09 (s, 3H), 2.53-2.60 (m, 2H), 2.66-2.73 (m, 1H), 2.77-2.86 (m, 1H), 2.89-2.95 (m, 1H), 3.09-3.15 (m, 1H), 3.72 (s, 3H), 4.12-4.13 (m, 1H), 7.80 (s, 2H), 11.91 (bs, 1H); 13 C NMR (DMSO-d₆), δ 22.45, 27.03, 40.41, 40.82, 53.87, 60.22, 90.86, 124.18, 139.29, 140.43, 143.70, 155.49, 156.72, 167.99; IR (KBr) 3428, 3252, 2914, 1675, 1636, 1565. Anal. ($C_{16}H_{17}N_3SO_2I_2$): C, H, N. 10 was obtained as a white powder by treating the above-obtained free base with maleic acid in CH₃CN: m.p. 121 °C (dec); 1 H NMR (CD $_{3}$ OD), δ 2.18 (s, 3H), 2.93-3.09 (m, 4H), 3.38-3.44 (m, 1H), 3.63-3.71 (m, 1H), 4.93-4.96 (m, 1H), 6.26 (s, 8/3H), 7.85 (s, 2H); ^{13}C NMR (DMSO-d₆), δ 22.40, 23.37, 37.73, 40.42, 53.73, 60.26, 91.59, 118.22, 134.71, 135.43, 140.67, 142.07, 157.45, 157.78, 167.07, 168.60; IR (KBr) 3436, 3049, 2968. 2946, 1700, 1686, 1624, 1571 cm⁻¹. Anal. ($C_{16}H_{17}N_3SO_2I_2 \cdot 4/3 C_4H_4O_4 \cdot 1/3 Et_2O$): C, H, N.

2-Amino-5-(2-phthalimidoethyl)thiazole Hydrobromide (27). In the same manner as **13**, the title compound was prepared from thiourea (2.75 g, 36.1 mmol) and **26**³² (crude, 10.69 g, 36.1 mmol). Recrystallization from MeOH/EtOH (1:10) gave 6.19 g (46% based on aldehyde **25**³²) of **27** as colorless plates: m.p. 244-246 °C (dec) (lit.³² 180 °C (dec starting point)); ¹H NMR (DMSO-d₆), δ 2.96 (t, J = 6.2 Hz, 2H), 3.77 (t, J = 6.3 Hz, 2H), 7.08 (s, 1H), 7.82-7.89 (m, 4H), 9.01 (bs, 2H); IR (KBr) 3308, 3222, 3108, 2975, 1764, 1711, 1618, 1608, 1554, 1402 cm⁻¹. Anal. (C₁₃H₁₂N₃O₂SBr): C, H, N.

N-2-(2'-Amino-5'-thiazolyl)ethyl-3,4,5-trimethoxyphenylacetamide (21), In the same manner as 17, the title compound was prepared from 20³² (1.95 g, 6.4 mmol) and 3,4,5-trimethoxyphenylacetyl chloride (1.57 g, 6.4 mmol). Recrystallization from CHCl₃/hexanes gave 1.42 g (63.3%) of 21 as colorless crystals: m.p. 121-122 °C; ¹H NMR (CDCl₃), δ 2.74 (t, J = 6.0 Hz, 2H), 3.34 (q, J = 6.3 Hz, 2H), 3.43 (s, 2H), 3.79 (d, 9H), 5.22 (bs, 2H), 5.78 (t, J = 5.6 Hz, 1H), 6.39 (s, 2H), 6.59 (s, 1H); ¹³C NMR (CDCl₃), δ 26.84, 40.39, 43.99, 56.07, 60.78, 106.35, 124.42, 130.29, 135.80, 137.09, 153.48, 167.34, 170.90; IR (KBr) 3294, 3114, 2995, 2936, 1654, 1636, 1588 cm⁻¹. Anal. (C₁₆H₂₁N₃SO₄): C, H, N.

N-2-(2'-Amino-5'-thiazolyl)ethyl-3,4,5-trimethoxyphenethylamine Dihydrochloride (24). To a suspension of 21(176 mg, 0.5 mmol) in dry THF (0.5 ml) was added dropwise slowly BH₃•THF(1.0 M in THF, 3.5 ml). After the addition was complete, the reaction mixture was stirred at room temperature for 30 min, and then was heated at reflux for 1 h. The reaction mixture was cooled to room temperature and treated cautiously with 10% HCl aqueous solution (3 ml), and the solution was heated at reflux for 30 min. After removal of THF, H₂O (10 ml) was added to the residue. The acidic solution was basified with 10% NaOH aqueous solution at 0 °C. The product was extracted with CHCI₃ (20, 20, 10 ml), the combined organic was washed with brine (20 ml), and dried over anhydrous Na₂SO₄. After filtration and evaporation, a viscous oil was obtained, it was dissolved in CHCl₃ (8 ml) and was treated with HCI (1.0 M in dry $\mathrm{Et_2O}$). The whole mixture was evaporated to dryness and the resulting white solid was crystallized in MeOH/Et₂O to give 109 mg (53.1%) of **24** as white crystals: m.p. 235 °C (dec); 1H NMR (CD $_3$ OD), δ 2.98-3.03 (m, 2H), 3.11-3.16 (m, 2H), 3.29-3.34(m, 4H), 3.72 (s, 3H), 3.84 (s, 6H), 6.63 (s, 2H), 7.18 (s, 1H); 13 C NMR (CD $_{3}$ OD), δ 24.63, 33.53, 48.38, 50.25, 56.73, 61.07, 107.24, 121.73, 125.27, 133.76, 138.28, 154.87, 172.01; IR (KBr) 3427, 2947, 2767, 1630, 1590 cm $^{-1}$. Anal. ($C_{16}H_{25}N_3SO_3CI_2$): C, H, N.

RADIOLIGAND BINDING STUDIES WITH β_1 ADRENORECEPTORS, β_2 ADRENORECEPTORS AND β_3 -ADRENORECEPTORS EXPRESSED IN CHO CELLS.

Competitive and comparative binding experiments on β_1 Adrenoreceptors, β_2 Adrenoreceptors and β_3 -Adrenoreceptors expressed in CHO cells were performed as described previously. Fraundorfer, P. F.; Fertel, R. H.; Miller, D. D.; Feller, D. R.

"Biochemical and pharmacological characterization of high-affinity trimetoquinol analogs on guinea pig and human beta adrenergic receptor subtypes: evidence for partial agonism." J Pharmacol Exp Ther 1994, 270, 665-74. ("Fraundorfer II", supra) CHO cells expressing human $β_1$ Adrenoreceptors, $β_2$ Adrenoreceptors and $β_3$ -Adrenoreceptors (provided by A. D. Strosberg, Institut Cochin de Genetique Moleculaire, Paris, France; and David Bylund, University of Nebraska, Omaha, NE; respectively) were cultured in Ham's F-12 medium supplemented with 10% fetal bovine serum, 50 U/mL-50 μg/mL of penicillin-streptomycin, 2 mM L-glutamine and 50 μg/mL of Geneticin in a humidified atmosphere of 5% CO₂-95% air. CHO cells grown to a confluence in 150-mL flasks were detached into Ham's F-12 medium after treatment with 0.05% trypsin-0.53 mM EDTA solution. The cells were then pelleted and washed three times with Tris-EDTA buffer (50 mM Tris-HCl, 150 mM NaCl, 20 mM EDTA, pH 7.4) and resuspended in the same buffer.

Data are expressed as the means \pm SE of the given number of experiments. All concentration-response and competition binding curves were analyzed using GraphPad Prism (GraphPad Software, San Diego, CA, USA). pK_{act} values are expressed relative to the maximal effect for each compound or effect at the highest concentration tested (for compounds with limited solubility). Relative efficacies (e_π) were calculated from plots of fractional percent occupancy versus response (% increase in cAMP accumulation) as described by Furchgott and Bursztyn (1967). The relative efficacies are expressed relative to (-)-isoproterenol, a reference $\mathfrak G$ -adrenoceptor agonist.

Competition binding experiments were performed in duplicates using these whole cells. Aliquots (150 μ L) of cells were added to tubes containing 50 μ L of [1251]ICYP (1.5-5 X 10⁴ cells/20-60 pM of ICYP) and varying concentrations of competing drugs. The final volume in each tube was 0.25 mL. Nonspecific binding (5-30%) was determined in the presence of 1 μ M (±)-propranolol. Incubations were carried out for 60 min at 37°C. Binding reactions were terminated by rapid filtration through Whatman GF/B glass fiber filters on a Brandel model 12-R tissue harvester. Filters were washed twice with ice cold Tris-EDTA buffer to remove free ICYP. The filters were dried under tissue harvester vacuum and radioactivity was measured by

gamma scintillation spectrometry (Beckman model 8000 gamma counter, Palo Alto, CA). Specific binding to β Adrenoreceptor sites in these cells varied from 94 to 100%

THROMBOXANE A₂ /PROSTAGLANDIN H₂ (TP) RECEPTOR SITES IN HUMAN PLATELETS.

For binding experiments, human platelet rich plasma (PRP) was centrifuged and re-suspended in 50 mM Tris-saline buffer, pH 7.4. Shin, Y.; Romstedt, K. J.; Miller, D. D.; Feller, D. R. "Interactions of nonprostanoid trimetoquinol analogs with thromboxane A₂/prostaglandin H₂ receptors in human platelets, rat vascular endothelial cells and rat vascular smooth muscle cells." J Pharmacol Exp Ther 1993, 267, 1017-23. Platelets were incubated with 5 nM [³H]SQ 29,548 in a final vol of 0.5 mL as described by Hedberg, A.; Hall, S.; Ogletree, M.; Harris, D.; Liu, E. "Characterization of [5-6³H]SQ 29,548 as a high affinity ligand for thromboxane A₂/prostaglandin H₂ receptors in human platelets". *J. Pharmacol. Exp. Ther.* **1988**, 245, 786-792. Unlabelled SQ 29,548 (50 μM) was used to determine nonspecific binding. Varying concentrations of each competing drug were used to quantify the inhibition of specific [3H]SQ 29,548 binding. Samples were incubated 30 min at room temperature, and rapidly filtered by vacuum through Whatman GF/C glass fiber filters on a Brandel cell harvester and washed for 10 sec with ice cold TRISsaline buffer. Filters were placed in plastic scintillation vials containing 10 mL of an emulsion-type scintillation mixture and radioactivity measured by liquid scintillation spectrometry. Specific binding to human platelets varied between 88 to 95%.

Competitive binding data were analyzed using the PC-version of the radioligand binding program LIGAND (McPherson, 1985). Inhibitory concentration-50 (IC50) value of each competing drug was determined graphically from individual plots of percent radioligand bound versus log drug concentration on β-adrenoceptors and human platelets. According to the reported method (Cheng, Y.; Prusoff, W. H. Relationship between the inhibition constant (K_i) and the concentration of the inhibitor which causes 50 percent inhibition (I₅₀) of an enzymatic reaction, *Biochem. Pharmacol.* **1973**, *22*, 3099-3108), K_i values were calculated from the obtained IC₅₀

values. Dissociation constants (Ki) for each competing drug were calculated using the equation:

$$K_i = \frac{IC_{50}}{(1 + \frac{L}{K_L})}$$

and the data expressed as pK_i (i.e., -log K_i) values. The K_L values used in the above equation are 17 pM, for β_1 Adrenoreceptor; 10 pM for β_2 Adrenoreceptor; 11 pM for β_3 -Adrenoreceptor; and 3.1 nM for Thromboxane A₂ /Prostaglandin H₂ receptors, respectively.

cAMP Radioimmunoassay (cAMP-RIA Assay). Chinese hamster ovary (CHO) cells expressing either human β_1 -, β_2 - or β_3 -adrenoceptor (AR) subtypes were used as previously described (Fraundorfer, P. F.; Lezama, E. J.; Salazar-Bookaman, M. M.; Fertel, R. H.; Miller, D. D.; Feller, D. R. Isomeric-activity ratios of trimetoguinol enantiomers on β-adrenergic receptor subtypes: functional and biochemical studies, Chirality 1994, 6, 76-85). These cells were grown to confluence in 60 mm dishes, washed with Hank's balanced salt solution, and then incubated with Hank's balanced salt solution (pH 7.4) containing 20 mM HEPES and 1 mM 3-isobutyl-1methylxanthine (IBMX) and 1 mM L-ascorbic acid for 30 min at 37 °C. Varying concentrations (10⁻¹¹ to 10⁻⁴ M) of the drugs were added with an additional 30 min of incubation. After removal of the Hank's buffer, the cAMP generated within the cells was extracted by the addition of trichloroacetic acid (6% w/v). cAMP content was determined as the amount of [125]-labeled succinyl-cAMP tyrosine methyl ester/antibody precipitated, as described by Brooker et al. (Brooker, G.; Harper, J. F.; Terasaki, W. L.; Moylan, R. D. Radioimmunoassay of cyclic AMP and cyclic GMP. In Advances in Cyclic Nucleotide Research; Brooker, G., Greengard, P. and Robinson, A., Ed.; Raven Press: New York, 1979, pp 1-33). The precipitated protein was dissolved in 0.1N NaOH. Protein content was determined by the method of Lowry et al. (Lowry, O. H.; Rosebrough, N. J.; Farr, A. L.; Randall, R. J. Protein measurement with the Folin phenol reagent., J. Biol. Chem. 1951, 193, 265-275), using bovine serum albumin as the standard. Data are expressed as the means ±SE of the given number of experiments. All concentration-response and

competition binding curves were analyzed using GraphPad Prism (GraphPad Software, San Diego, CA, USA). pK_{act} values are expressed relative to the maximal effect for each compound or effect at the highest concentration tested (for compounds with limited solubility). Relative efficacies (e_π) were calculated from plots of fractional percent occupancy versus response (% increase in cAMP accumulation) as described by Furchgott and Bursztyn (Furchgott, R.F., and Bursztyn, P. "Comparison of dissociation constants and relative efficacies of selected agonists on parasympathetic receptors" *Ann. N.Y. Acad. Sci.* **1967**, 882-889).

cAMP response element (CRE)-luciferase(LUC) reporter gene (CRE-LUC) assay. CHO cells stably expressing human β₁-, β₂-, or β₃-AR subtypes were transfected with a 6 CRE-LUC plasmid (gift from Dr. A. Himmler, Vienna, Austria) using electroporation with a single 70 ms, 150V pulse (Vansal, S. S.; Feller, D. R. Development of a rapid and efficient cyclic AMP assay for evaluating β-adrenergic receptor ligands., *Naunyn-Schmiedeberg's Arch. Pharmacol. Suppl.* 2 1998, 258, R659). The transfected CHO cells were seeded at a density of 40,000/well in 96 well microtiter plates (Culturplate, Packard) and allowed to grow for 20 hours. After 20 hours, the cells were treated with varying drug concentrations (10⁻¹¹ to 10⁻⁴ M) for 4 hours. Following drug exposures, the cells were lysed and luciferase activity measured using the LucLite® assay kit (Packard). Changes in light production were measured by a Topcount® luminometer (Packard).

FUNCTIONAL ACTIVITY OF TMQ ANALOGS IN ISOLATED RAT TISSUES

Male Sprague Dawley rats (Harlan Industries, Cumberland, IN) housed under a 12 hour light/dark cycle and fed Purina Rodent Laboratory Chow (Ralston Purina., St. Louis, MO) and water ad libitum, were used for the studies. On the day of the experiment, the rats weighting 200-430g were killed by cervical dislocation and tissues were quickly removed according to standard procedures (Staff of the Department of Pharmacology, University of Edinburg, in Pharmacological Experiments on Isolated Preparations, p. 104. Livingstone, London, 1968). Chronotopic responses of spontaneously beating right atria were used as a model for measuring \(\mathbb{G}_1\)-AR mediated activity (Konkar, A.A., Fraundorfer, P.F., Fertel, R.H., Burkman, A.M., Miller, D.D., and Feller, D.R. "Pharmacological Activities of trimetoquinol and 1-benzyl halogen-substituted analogues on rat β-adrenoceptor subtypes. Eur. J. Pharmacol. 1996, 305, 63-71). Relaxations of spirally cut tracheal strips precontracted with 3 x 10⁻⁷M carbachol, and of longitudinal segments of the esophageal smooth muscle precontracted with 10-6M carbachol (in the presence of 1 μM pindolol and 10 μM phentolamine), were used to measure β₂-and β₃-ARmediated activity, respectively (Konkar, A.A., Fraundorfer, P.F., Fertel, R.H., Burkman, A.M., Miller, D.D., and Feller, D.R. "Pharmacological Activities of trimetoquinol and 1-benzyl halogen-substituted analogues on rat β-adrenoceptor subtypes. Eur. J. Pharmacol. 1996, 305, 63-71; Lezama, E.J., Konkar, A.A., Salazaar-Bookaman, M.M., Miller, D.D., and Feller, D.R. "Pharmacological study of atypical β-adrenoceptors in rat esophageal smooth muscle. Eur. J. Pharmacol., 1996, 308, 69-80). Contractions of spirally cut aortal strips and inhibition of phenylephrine-induced contraction of the tissue, were used to measure α-AR mediated agonist or antagonist activity of the compounds respectively. The tissues were isolated and prepared for measurement of functional activity as per protocols described earlier (Konkar, A.A., Fraundorfer, P.F., Fertel, R.H., Burkman, A.M., Miller, D.D., and Feller, D.R. "Pharmacological Activities of trimetoquinol and 1benzyl halogen-substituted analogues on rat β-adrenoceptor subtypes. Eur. J. Pharmacol. 1996, 305, 63-71; Staff of the Department of Pharmacology, University

of Edinburg, in Pharmacological Experiments on Isolated Preparations, p. 104. Livingstone, London, 1968). All tissues were suspended and equilibrated in modified Kreb's buffer in water-jacketed baths at 37°C. Resting tensions of 1 g for right atria, trachea and aorta, and of 200 mg for esophageal smooth muscle were used. All tissue responses were measured on a Grass Polygraph Model 7C with a Grass FT-03C isometric force-displacement transducer. Cumulative concentration response curves for each drug were constructed by the method of van Rossum (van Rossum, J.M. "Cumulative dose-response curves. II. Technique for making of dose-response curves in isolated organs and the evaluation of drug parameters. *Arch. Int. Pharmacodyn.* 1963, 143, 299-300). Increasing concentrations of compounds were added every 2-3 min with

(-)-isoproterenol and every 10-15 min with TMQ analogs, or until no further change in response was observed.

In the right atrium, the concentration response curve for (-)-isoproterenol was followed by complete washout of the drug, after which a curve for either acetamido- or chloroacetamido DITMQ was constructed. The tissue was again washed 6-7 times with 10 ml of buffer, followed by repeated washes every 10-15 min. Changes observed in the duration of chronotropic effect following repeated washes was used as an indicator of 'irreversible' binding of the compound to the atrial tissue. A second concentration response curve with (-)-isoproterenol was constructed in atria to determine desensitization effects.

The concentration response curves in trachea and esophagus culminated with a final concentration of 10⁻⁵M(-)-isoproterenol, to determine the maximal relations induced in the tissue and express functional responses of the TMQ analogs as a percentage of maximal

(-)-isoproterenol-induced response. Carbachol-precontracted tissues were included as controls through the duration of relaxation studies.

Studies with aorta were performed in the presence of 1 μ m pindolol,to block ß-AR-mediated effects. Concentration response curves to phenylephrine were followed by washout of the drug 30 min incubation with 10⁻⁵ M acetamido- or chloroacetamido-DITMQ. A second phenylephrine concentration-response was then

constructed to determine any α -AR blocking activity of the TMQ analogs. In control experiments, second concentration response curves of phenylephrine were constructed in the absence of treatment with the compounds.

aUsing [¹²⁵I] ICYP as radioligand, N=4-9
bUsing [³H] SQ 29,548 as radioligand, N=4-9
cPR = potency ratio relative to cmpd. 1 (TMQ). PR = antilog [pKi(drug)-pKi(TMQ)]

Table 2. Selectivity of Trimetoquinol (TMQ) Analogs for Human $\beta 2\text{-}$ and $\beta 4\text{-}$

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	pKi	pKi ± SEM	
Compound	human β1 CHOa	human β2 CHOa	β2/β1 selectivityb
-	6.49±0.06	7.36±0.23	7.4
2	7.10±0.06	8.69±0.16	39
21a	6.74±0.30	9.52 ± 0.13	009
211-11-110511	11. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1.		

aUsing [125 I] ICYP as radioligand for β_1 - and β_2 Adrenoreceptor expressed in CHO cells, N=4-9 $b_{\beta 2/\beta 1}$ -selectivity = Ki (β_1 Adrenoreceptor)/Ki (β_2 Adrenoreceptor) Table 3. Agonist Activities of (-)-Isoproterenol, AcetamidoDITMQ and Chloracetamido-DITMQ on ß-Adrenoceptors in Isolated Rat Tissues

Data are calculated as pEC _s response elected by the drug response elicited by (-)-isopr parentheses. † Indicates signisoproterenol (P<0.05).	Data are calculated as pEC ₅₀ (-log EC ₅₀ , concentration required to produce a response equal to 50% of maximal response elected by the drug) and I.A. (Intrinsic activity, maximal drug-induced response relative to the maximal response elicited by (-)-isoproterenol). The values are mean ± SEM of the number of experiments indicated in parentheses. † Indicates significant difference in value of TMQ analog compared to corresponding value of (-)-isoproterenol (P<0.05).	ed to produce a response edimal drug-induced response ± SEM of the number of exi 10 analog compared to corr	qual to 50% of maximal relative to the maximal seriments indicated in esponding value of (-)-
	Right Atria (ß-AR)	Trachea	Esophageal Smooth Muscle (Atypical-ß/ß ₃ -AR)
(-)Isoproterenol			
pECso	8.95 ± 0.06	8.00 ± 0.05	7.34 ± 0.08
I.A.	1.00	1.00	1.00
(u)	(5)	(8)	(12)
AcetamidoDITMQ (A-14)			
pECso	8.96 ± 0.04	9.22 ± 0.07†	8.68 ± 0.12
I.A.	0.93 ± 0.04	0.84 ± 0.02†	0.99 ± 0.03†
(u)	(4)	(4)	(7)
ChloroacetamidoDITMQ (A-37)			
pECs	8.94 ± 0.07	8.90 ± 0.05†	8.08 ± 0.03†
I.A.	0.81 ± 0.05 †	0.83 ± 0.02 †	0.99 ± 0.01
(u)	(4)	(4)	(9)

Table 4. Inhibition constants (-log K ₁ or pK ₁) for binding and functional activity constants (-log EC ₅₀ or pK _{3c1}) for cAMP accumulation in CHO cells expressing rat-8 ₃ -AR. E _{max} is the maximal cAMP accumulation stimulated by the	constants (-log K ₁ or pK ₁) for binding and functional activity constants (-log EC ₅₀ or pK _{3cl}) for n in CHO cells expressing rat-8 ₃ -AR. E _{max} is the maximal cAMP accumulation stimulated by	onal activity constants (-los maximal cAMP accumul	og EC _{so} or pK _{act}) for lation stimulated by the
in parentheses. Ind = value not determined. It indicates significant difference in value of TMQ analog compared to corresponding to value of (-)-isoproterenol (P<0.05).	 a value not determined. I Indicates significant difference in value of TMQ analog compared value of (-)-isoproterenol (P<0.05). 	t difference in value of TM	xperiments indicated
COMPOUND	ላ ሚ	PK _{ad}	Щ жа
(-)Isoproterenol	4.45 ± 0.06 (14)	7.90 ± 0.12 (16)	100
BRL 37344	6.96 ± 0.08 (5) 1	8.90 ± 0.12 (9) 1	103.3 ± 6.80
S(-)TMQ	5.67 ± 0.03 (5) 1	8.19 ± 0.19 (7)	87.65 ± 6.90
(±)TMQ	5.11 ± 0.12 (8) 1	8.69 ± 0.14 (10) 1	125.3 ± 9.20
DITMQ (A-11)	6.34 ± 0.03 (5) 1	9.40 ± 0.08 (9) 1	96.90 ± 10.10
AminoDITMQ (A-35)	6.14 ± 0.08 (5) 1	pu	pu
AcetamidoDITMQ (A-14)	7.28 ± 0.06 (5) 1	9.34 ± 0.12 (6) 1	89.57 ± 6.93
ChloroacetamidoDITMQ (A-37)	6.49 ± 0.05 (5) 1	9.05 ± 0.16 (6) 1	92.48 ± 6.59
BromoacetamidoDITMQ (A-38)	6.70 ± 0.05 (6) 1	9.36 ± 0.39 (5) 1	87.27 ± 12.21
6,7-Dimethoxy-acetamidoDITMQ	4.84 ± 0.03 (5) 1	pu	pu
6,7-Dimethoxy TMQ	3.88 ± 0.07 (4) 1	no activity upto 3 x 10 ⁻⁵ M	M
6,7-Methylenedioxy TMQ	4.29 ± 0.05 (5)	5.92 ± 0.12 (8) 1	70.90 ± 4.411*
DemethoxyDITMQ	5.80 ± 0.03 (5) 1	8.74 ± 0.10 (4) 1	97.69 ± 5.85
IsothiocyanatoITMQ (A-46)	5.83 ± 0.12 (4) 1	8.61 ± 0.15 (4) t	104.1 ± 7.10

 a Human $\beta_{1^-},\,\beta_{2^-}$ and $\beta_{3^-}AR$ were expressed in CHO cells. [126 JICYP was used as the radioligand. K_i values were calculated using the following equation: K_i (nM) = IC $_{50}$ $(1 + [L]/K_d)$, wherein IC_{so} is the concentration (nM) of an analog at which the radioligand binding was reduced by 50%; [L] is the radioligand concentration used; K_d is the radioligand equilibrium dissociation constant. $pK_i = -logK_i$; SEM = standard error of mean. N = 3-9. Table 5. Human β-Adrenoceptors Binding Affinities of TMQ Analogs

	pK,±	pK,± SEMª	
	Human β,-AR	Human β ₂ -AR	Human B ₃ -AR
OSI	5.80 ± 0.07	6.17 ± 0.12	4.73 ± 0.25
TMQ	6.49 ± 0.06	7.36 ± 0.23	5.43 ± 0.28
8 (A-11)	7.10 ± 0.06	8.69 ± 0.16	7.67 ± 0.24
7 (B-29)	5.21 ± 0.08	6.21 ± 0.12	4.17 ± 0.05
9 (B-28)	6.14 ± 0.08	6.37 ± 0.08	5.83 ± 0.15

^a Human $β_1$, $β_2$ - and $β_3$ -AR were expressed in CHO cells. K_{act} is the molar drug concentration which produces a cAMP response equal to 50% of its maximal response, $pK_{act} = -logK_{act}$. ^b I.A. = Intrinsic Activity, expressed as the percentage of a maximal analog response relative to the maximal response (100%) of R-(-)-isoproterenol (ISO). ^c see Experimental Section. ^d Data is for S-(-)-TMQ isomer. ^e N.A. = Not active at 100 μM. Values are the mean ± SEM of N = 4-12. Table 6. Human β-Adrenoceptors (AR) Functional Activities of TMQ Analogs

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	пишаг	numan Is 1-AK	Humar	Human BAR	Himar	Human RAR
	pK _{act} ± SEMª	I.A. ± SEM⁵	pK _{act} ± SEM	I.A. ± SEM	pK _{act} ± SEM	377
		A.	A. CAMP-KIA ASSAY	<u>\</u> رد		
osl	8.75 ± 0.14	100	8.40 ± 0.17	100	7 37 + 0 11	700
TMO	870 +011	400	0000		1.0 + (6.7	001
3		01 H 601	8.33 ± 0.24	95 + 3	8.60 ± 0.15	95 + 3
8 (A-11)	8.11 ± 0.13	103±4	8.47 ± 0.12	56 + 9	8 76 + 0.2	120 + 0
7 (R-20)	9 4 14	0,7			2.0 - 0.10	120 I 9
(0.40)	ζ.	< 10	Z.A.	20 ± 1	5.06 ± 0.01	54 + 1
9 (B-28)	Z.A.	< 5	N.A.	<.5	6 95 + 0 11	67 - 2
10	< 2	7.			0.30	0/ H 3
2	.Y.Y.	01.>	N.A.	<10	A.N	< 10
		æi	B. CRE - LUC assave	JA _C		
9 (R-28)	Ø Z		*	1		
0 (0-50)		\ \ \	Ä.Ä	< 15	6.71 ± 0.18	62 + 3
24	N.B.	< 10	A.N	< 10	V N	7,
				,	֡֝֝֝ <u>֚</u>	2 ~

Data on TMQ analogs using the CRE-LUC assay

Compound β₁-AR β₂-AR β₃-AR pK _{act} ±SEM I.A. pk _{act} ±SEM I.A. pK _{act} ±SEM I.A. pK _{act} ±SEM (n) (n) (n) 6-Monophenolic TMQ 5.71±0.13 58 6.64±0.15 38 7.45±0.09 90 analog (A-4) (8) (8) (8) (8) (8) (8) (8) Analog (A-3) (8) (8) (8) (8) (8) (8)	Table 7. Agonist potencies (pKact values) and intrinsic activities (IA) of noncatechol (A 3) and 6-monophenolic (A 4) analogs for human β-adrenoceptor subtypes expressed in Chinese hamster ovary cells. IA values are expressed relative to the maximal response to (-)-isoproterenol in these CHO cell systems expressing the three human β-adrenoceptor subtypes. Agonist activities were measured using the cyclic AMP response element-luciferase reporter gene (CRE-LUC) assay.	incies (pKact variolic (A 4) anald hamster ovary -)-isoprotereno ceptor subtype element-lucifera	alues) and alues) and alues) and alues for cells. I in the series are repaired and alues are alu	and intrinsic a human β-adra IA values are se CHO cell s nnist activities orter gene (Cl	ctivities enocepto express ystems were me	(IA) of noncator subtypes ed relative to expressing the easured using assay.	echol the e
1.A. pk _{act} ±SEM 1.A. pK _{act} ±SEM (n) (n) (n) (n) (n) 22 N.D. 21 7.55±0.22 (8) (8)	Compound	β ₁ -AR		β ₂ -AF	~	B ₂ -AR	
58 6.64±0.15 38 7.45±0.09 (8) (8) (8) (8) (8) (8) (8) (8) (8) (8)		pK _{act} ±SEM (n)	I.A.	pk _{act} ±SEM		pK _{act} ±SEM	<u>4</u>
N.D. 22 N.D. 21 7.54±0.22 (8)	6-Monophenolic TMQ analog (A-4)	5.71±0.13 (8)	58	6.64±0.15	38	7.45±0.09	06
	Non-catechol TMQ Analog (A-3)	N.D.	22	N.D.	21	7.54±0.22	62

ors Expressed in ing Human B ₃	hß ₂ Binding ki (nM)	534 +/- 67	24 +/- 9	3.1 +/- 1.7	
d ß, Adrenorecept	hß, Binding KI (nM)	8400 +/- 8100	14.2 +/- 9.8	25 +/- 1.3	63.4 +/- 0.9
for Human ß ₁ , ß ₂ a of TMQ Analogs in	h ß, cAMP % Iso +/- SEM	32.3 +/- 5	81 +/- 3	119 +/- 6	108 +/- 2
of TMQ Analogs fo	hß, cAMP EC _{so} (nM) +/- SEM	506 +/- 137	45 +/- 9	45	4
Table 8 Selectivity of TMQ Analogs for Human Ω_1 , Ω_2 and Ω_3 Adrenoreceptors Expressed in CHO Cells and Functional Activities of TMQ Analogs in CHO Cells Expressing Human Ω_3 Adrenoreceptors.	MOLECULAR STRUCTURE	HN S N'tH	HOOH	HOO NH	HOOH HOO

	Т -			
otors Expressed in sing Human $ extbf{\mathbb{B}}_3$	hß _z Binding ki (nM)	>3000	47 +/- 4.3	6.5 +/- 1.5
nd 8, Adrenorecer	hß, Binding KI (nM)	>3000	278 +/- 59	6.7 +/- 1.3
Table 8 Selectivity of TMQ Analogs for Human \mathcal{B}_1 , \mathcal{B}_2 and \mathcal{B}_3 Adrenoreceptors Expressed in CHO Cells and Functional Activities of TMQ Analogs in CHO Cells Expressing Human \mathcal{B}_3 Adrenoreceptors.	h ß ₃ cAMP % Iso +/- SEM		89 +/- 5	100 +/- 2
	hß ₃ cAMP EC ₅₀ (nM) +/- SEM	>3160	33.5 +/- 7	₹
Table 8 Selectivity CHO Cells and Fun Adrenoreceptors.	MOLECULAR STRUCTURE	wh ₂ / NH· HCI	HO NH·HCI	HO NH HO OMe OMe OMe

CONCLUSION

The invention has been described herein with regard to particular preferred operating circumstances and requirements, and in a particular context. Those of ordinary skill will clearly understand the application of the invention and its uses in other diverse circumstances and will, with the guidance provided herein, be able to adapt the invention to the particular requirements of other contexts of practice of the invention.

The foregoing description and disclosure of the present invention is intended to be illustrative for the guidance of those of ordinary skill in the art to which the invention pertains, and is not intended to define or limit the scope of the invention. The scope of the invention is defined and limited only in the appended claims.